

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

	CANDIDATE NAME		
	CENTRE NUMBER		CANDIDATE NUMBER
*			
2	CO-ORDINATED SCIENCES		0654/23
3	Paper 2 (Core)		October/November 2013
			2 hours
4 6	Candidates ans	wer on the Question Paper.	
2 4	No Additional M	aterials are required.	
4			

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units. A copy of the Periodic Table is printed on page 28.

At the end of the examination, fasten all your work securely together. The number of marks is given in brackets [] at the end of each question or part question.

This document consists of 28 printed pages.



[Turn over

1 Fig. 1.1 shows a root hair cell.



- Fig. 1.1
- (a) Use the letters A, B and C to label these parts of the root hair cell in Fig. 1.1.
 - **A** the cell membrane
 - B the part that contains chromosomes
 - C a structure that is not present in animal cells

[3]

For

- (b) Name two substances that are absorbed by root hair cells.
 - 1 _____ 2 _____

[2]

- xylem phloem Fig. 1.2 (i) State the function of phloem. (ii) Suggest why this treatment would cause the roots of the plant to die.
- (c) Fig. 1.2 shows part of a plant stem from which the outer layer, including the phloem, has been removed.

2 (a) Table 2.1 shows information about some chemical elements and their positions in the Periodic Table.

element group number in the Periodic Table 6 oxygen calcium 2 lithium 1 sulfur 6 7 fluorine (i) State the noble (inert) gas that is in the same period of the Periodic Table as sulfur. [1] (ii) Select two elements from Table 2.1 whose atoms form ionic chemical bonds with each other and explain your answer. and explanation [2] (b) Fig. 2.1 shows a diagram of an atom. Α Fig. 2.1

Table 2.1

(i) Name structure A in Fig. 2.1. [1]

(ii)	State the proton number of the atom in Fig. 2.1.	For
	Explain your answer briefly.	Use
	proton number	
	explanation	
	[2]	
(c) A ma	student added excess acidified barium chloride solution to a solution of a gnesium compound to produce mixture W .	
Fig	. 2.2 shows the procedure followed.	
acidified ba	rium	
$\left\{ \right.$		
solu mag com	tion of a nesium pound white precipitate settled on the bottom of the beaker	
	Fig. 2.2	
(i)	Suggest the full name of the magnesium compound in the original solution.	
	[1]	
(ii)	Describe briefly what the student should do to find the mass of the white precipitate in mixture ${f W}.$	
	[3]	

3 (a) Fig. 3.1 shows a circuit used to measure the current passing through a resistor when the voltage across it is changed.

For Examiner's Use



Fig. 3.1

 (b) Complete the sentences below using a word or phrase from the list. Each word or For phrase can be used once, more than once or not at all. Examiner's Use decreases increases stays the same is zero When the voltage across the resistor is reduced, the current through the resistor When the voltage of the supply is reduced, the voltage across the resistor When the voltage across the resistor is reduced, the resistance of the wire . [2] (c) The resistance of a piece of wire depends on a number of variables such as the temperature of the wire and the material from which it is made. State two other factors which affect the resistance of a piece of wire. 1

2	[2]

Soya beans are an important crop in Brazil. Soya beans contain a lot of protein, plus 4 smaller quantities of starch and fat. Examiner's (a) Describe how you could test a sample of soya beans to find out if they contain fat. [3] (b) Explain why protein is an important part of a balanced diet. [2] (c) When a person eats soya beans, the beans are chewed in the mouth. Explain why this makes it easier for enzymes in the digestive system to digest the beans. [2] (d) Raw soya beans contain substances that stop protease enzymes from working. Cooking destroys these substances. Suggest how eating uncooked soya beans could prevent the absorption of some of the nutrients from them. [2]

8

For

Use

(e) Large areas of rainforest have been cleared in Brazil, to provide more land for growing soya beans. Examiner's

Explain how cutting down the rainforest can harm the environment.

..... [4] For

Use

5 (a) A student placed four equally-sized pieces of different metals into colourless liquids contained in four test-tubes **P**, **Q**, **R** and **S**.

Fig. 5.1 shows what the student observed.



10

For

Examiner's Use (b) In the process of copper plating, a thin layer of copper is formed on the surface of a metal object.

For Examiner's Use

Fig. 5.2 shows the apparatus and materials that are needed to copper plate a metal key.



Fig. 5.2

Draw a diagram which shows how the apparatus and materials in Fig. 5.2 should be assembled so that the metal key will be copper plated.

[3]

6 (a) Fig. 6.1 gives information about the uses of different types of electromagnetic waves and their effects on living tissue.

12

For Examiner's Use

Draw lines to link each electromagnetic wave with its effect on living tissue and its use. One has been completed as an example.





[4]

(b) Electromagnetic waves are transverse waves. Water waves are also transverse.

Draw a diagram of a transverse wave on the axes below. Label the amplitude and **one** wavelength on your diagram.

[3]

Please turn over for Question 7.

7 Ball pythons (royal pythons) are snakes that are kept as pets in many parts of the world.



The colour of a ball python is determined by its genes.

Some ball pythons are albino (white). This is caused by a recessive allele, **a**. The dominant allele, **A**, gives normal colouring.

(a) Complete Table 7.1 to show the possible genotypes and colours arising from this gene.

genotype	colour
AA	
Aa	normal
	albino

Table 7.1

[2]

For Examiner's Use

(b) State the correct biological term for the visible appearance produced by the genotype, in this case the colour of the snake.

[1]

(c) (i) Complete the genetic diagram to explain the results of crossing two snakes that are heterozygous for these alleles.



For

Examiner's Use 8 (a) Fig. 8.1 shows apparatus a student used to investigate the reaction between dilute nitric acid and excess calcium carbonate.

For Examiner's Use



Fig. 8.1 (i) Name the gas that is given off in this reaction.[1] (ii) Describe how the student could test for the gas you named in (i). You may wish to complete the diagram in Fig. 8.1 to help you to answer this question. [2] (iii) At the end of the reaction the test-tube in Fig. 8.1 contains a solution of the compound calcium nitrate. State the general name for compounds like calcium nitrate which are produced when an acid reacts with a metal carbonate.[1] (iv) The chemical formula of calcium nitrate is Ca(NO₃)₂. State the total number of atoms and the number of different elements that are shown combined together in this formula. total number of atoms

number of different elements	

[2]

(b) The student then carried out an investigation into the way that the rate of the reaction in(a) changed when he varied the concentration of the nitric acid.

Fig. 8.2 shows the apparatus the student used to measure the rate of reaction.



Fig. 8.2

The student measured the rate of reaction by finding how long it took for the gas syringe to fill with gas.

(i) After he had completed several measurements, the student wrote the following correct conclusion in his notebook.

Conclusion
The higher the pH of the dilute nitric acid
the longer it took for the gas syringe to
fill with gas.

Explain this conclusion briefly.

[2]

(ii) State **two** other variables that can affect the rate of reaction between dilute nitric acid and calcium carbonate.

1	
2	 [2]

17

For

Examiner's Use **9** Fig. 9.1 shows a solar-powered golf cart used to carry golfers around a golf course.



Fig. 9.1

- (a) As the cart moves around the course, the motion of the cart is measured.
 - Fig. 9.2 shows a distance/time graph for a small part of the journey lasting 60 seconds.



	(ii)	Calculate the speed of the cart between B and C .	For For
		Show your working.	Use
		m/s [1]	
	(iii)	Describe the motion of the cart between D and E .	
		[1]	
	(iv)	During another part of the journey, the cart is accelerating.	
		State whether the forces acting on the cart are balanced or unbalanced.	
		Explain your answer.	
		[1]	
(b)	The use	e cart is powered by solar cells on its roof. The solar cells produce electrical energy d to charge the rechargeable batteries in the cart.	
	Nar	ne one other renewable energy resource that could produce electrical energy.	
		[1]	
(c)	The	golfer hits a golf ball with his club. The ball flies through the air.	
	(i)	State the form of energy given to the golf ball when the ball is hit.	
		[1]	
	(ii)	State the form of energy gained by the golf ball as it rises into the air after being hit.	
		[1]	

(d) The mass of a golf ball is 45 g. The volume of a golf ball is 36 cm^3 .

Calculate the density of the golf ball.

State the formula that you use and show your working.

formula

working

_____ g/cm³ [2]

(e) (i) The head of the golf club is made of solid metal. The air that the golf ball is travelling through is a gas.

Complete Fig. 9.3 below to show the arrangement of particles in a gas. The diagram for a solid has been done for you.





[2]

For Examiner's Use

(ii) During the cart's journey, the temperature of the air in the tyres increases by 15 °C.

The volume of the air in the tyre remains the same.

Explain in terms of particles why the **pressure** of the air in the tyre increases when this happens.

[1]

(iii)	Sometimes the golfer's hands begin to sweat.	For
	Explain in terms of particles how sweating cools his hands.	Use
	[3]	

(iv) Suggest how this will affect the rate of diffusion of carbon dioxide from the blood to the alveoli. Examiner's

Explain your answer.	
effect on rate of diffusion	
explanation	
[2]

For

Use

Pet	roleu	um (crude oil) is a liquid fossil fuel.	For
(a)	Nar	me one solid fossil fuel. [1]	Use
(b)	Ga	soline and diesel are mixtures of liquid hydrocarbons obtained from petroleum.	
	(i)	Name the process used to separate gasoline and diesel from petroleum.	
	(ii)	State the main use of gasoline and explain, in terms of its chemical properties, why it is suitable for this use.	
		use	
		explanation	
		[2]	
(c)	Nat con	tural gas is a gaseous fossil fuel, which contains mainly methane mixed with other npounds such as ethane.	
	(i)	Complete the diagram of the structure of one molecule of ethane.	
		—C	
	(ii)	Complete the word chemical equation for the complete combustion of ethane	
et	hane		
		[2]	
(d)	Eth Eth cata	ene, C_2H_4 , is an unsaturated hydrocarbon. ene is manufactured by heating large hydrocarbon molecules in the presence of a alyst. During this process no air must be allowed into the reaction vessel.	
	(i)	Name the process used to manufacture ethene. [1]	

(ii) Suggest one reason why air must be kept out of the reaction vessel.

[2]

12 (a) Fig. 12.1 shows a light ray entering an optical fibre.



Fig. 12.1

The light ray travels all the way through the optical fibre.

Explain why the light ray is able to stay inside the optical fibre.

You may draw on the diagram if it helps your answer.

[2]

(b) White light is passed through a prism as shown in Fig. 12.2.



Fig. 12.2

- (i) State the colours seen at positions X and Y.
 - X ______ Y _____ [2]
- (ii) A rainbow is formed in a similar way. Suggest what is acting as a prism when forming a rainbow.

[1]

- (c) Fig. 12.3 shows a person looking into a mirror and seeing an image.
 - mirror
 - Fig. 12.3
 - (i) Write the letter X on Fig. 12.3 to show the position of the image of the person's nose.
 [2]
 - (ii) Select three words or phrases from the list that describe the image correctly.

larger than object	real	same size as object	
smaller than object	upright	upside down	virtual
			[3]

For

Examiner's Use

	III IV V VI VII 0	4	He	Helium 2	11 12 14 16 19 20	B C N O F	soron Carbon Nitrogen Oxygen Fluorine Neon 6 7 8 9 9 10	27 28 31 32 36.5 40	Al Si P S Cl Ar	minium Silicon Phosphorus Sultur Chlorine Argon 14 15 16 16 17 18	70 73 75 79 80 84	Ga Ge As Se Br Kr	allum Germanium Arsenic Selenium Bromine Krypton 32 33 34 35 36	115 119 122 128 127 131	In Sn Sb Te Xe	odum Tin Antimony Tellurium lodine Xenon 50 51 52 53 54	204 207 209	T <i>l</i> Pb Bi Po At Rn	alium Lead Bismuth Polonium Astatine Radon 82 83 84 85 86 86				162 165 167 169 173 175	Dy Ho Er Tm Yb Lu	prosium Holmium Erbium Thulium Ytterbium Lutetium 67 68 69 70 71		Cf Es Fm Md No Lr	_
-	=	-			1	8	5 Borc	27	A	Alumir 13	55 70	ů ľ	inc Galliu 31	12 115	i lr	Indiu Indiu 49	01 20	Hg T	rcury Thalli 81				59 162	ے م	bium Dyspro		S SK	
											64	Cu	Copper 30	108	Ag	Silver Cao	197 2	Au	Gold Re 80				157 1	C Cq	Sadolinium Ter		E C W	
dh											59	ïZ	Nickel 29	106	Pd	Palladium 47	195	Ł	Platinum 79 79				152	Eu	Europium 64		Am	
5											59	ပိ	Cobalt 27	103	Rh	Rhodium 45	192	<u>-</u>	Iridium 77				150	Sm	Samarium 62		Pu	-
		-	I	Hydrogen 1							56	Fe	Iron 26	101	Ru	Ruthenium 44	190	os	Osmium 76					Pm	Promethium 61		dN	-
											55	ЧШ	Manganese 25		۲	Technetium 43	186	Re	Rhenium 75				144	PN	Neodymium 60	238	D	
											52	ບັ	Chromium 24	96	Mo	Molybdenum 42	184	≥	Tungsten 74				141	ሻ	Praseodymium 59		Ра	
											51	>	Vanadium 23	93	qN	Niobium 41	181	Та	Tantalum 73				140	မီ	Cerium 58	232	Ч	_
											48	F	Titanium 22	91	Zr	Zirconium 40	178	Ŧ	+ Hafnium 72				1			mic mass	lodr	
-					[45	Sc	Scandium 21	89	≻	Yttrium 39	139	La	Lanthanum 57 *	227	Ac	Actinium 89 1	d series		Selles	<pre>i = relative ato.</pre>	A = atomic syn A	
-	=				6	Be	Beryllium 4	24	Mg	Magnesium 12	40	Ca	Calcium 20	88	Sr	Strontium 38	137	Ba	Barium 56	226	Ra	Radium 88	anthanoi			0 10	×	
	_				7	:-	Lithium	23	Na	Sodium 1	39	¥	Potassium 9	85	Rb	Rubidium 7	133	Cs	Caesium 5	1	Ľ	Francium 7	8-71		-10		ey	,

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