MARK SCHEME for the October/November 2009 question paper

for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/02

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Page	2	Mark Scheme: Teachers' version IGCSE – October/November 2009	Syllabus 0654	Paper 02
1	(a) (i)) (eleo	ctricity into) heat ;	0004	[1]
	(ii)) (eleo	ctricity into) kinetic energy / electricity into movement ;		[1]
	(b) he	eat :			
	hi	gh; ction;			[3]
					[Total: 5]
2	(a) 27	7.8 % ;			[1]
	re e.	moval g. silico	n is addition of oxygen / bonding with oxygen / reductio of oxygen ; on (dioxide) is reduced because oxygen is removed / c because it joins with oxygen ;		[2]
			, , , , , , , , , , , , , , , , , , ,		
	(c) 		nthode	anode electrolyte	
	al	•••••	ct = 2 marks, 2 correct = 1 mark		;; [2]
	(d) (i)	wea	thering / erosion / transportation ;		[1]
	(ii)	collo	bid / sol ;		[1]
	(iii)	heat	ted to a high temperature / it is fired ;		[1]
					[Total: 8]
3	(a) (i)		l source / energy source / nutrients ; embryo / for germination ;		[2]
	(ii)		ein - growth / repair ; ch - energy ;		[2]
	(iii)		biuret reagent / add copper sulfate and potassium hyd ole colour indicates protein ;	roxide (solution) ;	[2]
	(b) (i)	Dun	field ;		[1]
	(ii)	Man	idarin ;		[1]

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		(iii)	(mor	re) photosynthesis / CO_2 is a limiting factor ;	[1]				
	re		ref to	carbon dioxide in the atmosphere is increasing ; ref to a reason for this, e.g. burning fossil fuels / deforestation ;					
	idea of needing to plan for future food productio					[max 2]			
						[Total: 11]			
4	(a)	(i)	5;			[1]			
		(ii)	15 e	sphorus / P ; lectrons so 15 protons so atomic number 15 / ectrons in outer shell / in group 5 and three shells / per	iod 3 ;	[2]			
	(b)	(i)	nitro	gen / N ;		[1]			
		(ii)	-	together / form chains / form polymers ; eins / polypeptides ;		[2]			
	(c)	(i)	diffe	tion tent contains one type of atom but compound contains rent atoms (bonded) ; N_2 is an element and NH_3 is compound ;	;	[max 2]			
		(ii)	turns OR mix v	np), red litmus (paper) ; (allow universal indicator) s blue ; with HC <i>l</i> gas ;		[0]			
			uens	se white smoke ;		[2] [Total: 10]			
5	(a)	(i)	carb	on dioxide ;		[1]			
		(ii)		water ; s cloudy ;		[2]			
	(b)	(i)		sity = mass / volume ; /5.6 = 2.7 g / cm³ ;		[2]			
		(ii)		 particles touching, regular arrangement ; most particles touching, irregular arrangement ; 		[2]			
	(c)		erweig ect on						
		too exp		[max 2]					

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	(d) (i) tem			perature rise proportional to energy input ;		[1]		
		(ii)	work 40 0			[2]		
		(iii)	[1]					
		(iv)	v) power = energy / time ; 40 000 / 600 = 66.7 W ; (allow ecf)					
		(v)	<pre>(v) current = 66.7 / 12 = 5.5 A ; so fuse will not melt ;</pre>					
	(e)	(i)	beta alpha	; a would be completely stopped and gamma not stoppe	ed at all ;	[2]		
		(ii)	lead	;		[1]		
						[Total: 20]		
6	(a)	B D						
		C Any	/ two (correct for one mark ;;		[2]		
	(h)	(i)	cont	raata / aata chartar :				
	(0)	(i)		racts / gets shorter ; a ulna closer to, bone B / humerus / bone A , scapula ;		[2]		
		(ii)	trans	smit force from muscle to bone ;		[1]		
	(c)	(i)	arter capil	y; llary;		[2]		
		(ii)	-	thing rate / breathing depth, increases ;		[-]		
		()		t rate increases ;		[2]		
						[Total: 9]		
7	(a)	(i)	fract	ional distillation / fractionation ;		[1]		
		(ii)	lowe lowe lowe less	oline has: r viscosity ; r boiling point / more volatile ; r melting point ; coloured ;				
			highe	er flammability ;		[max 2]		

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	(iii)	mad orga	[2]				
	(b) (i)	(cata	alytic / thermal) cracking / thermal decomposition ;	[1]			
	• •		alkanes paired with saturated ; alkenes paired with unsaturated ;				
	(iii)	two	ne molecules contain a double bond ; carbon atoms required at either end of the double bon w diagram of second point)	d / owtte ;	[2]		
	(c) sulf whi (su whi dar	ic life /	[max 3]				
					[Total : 13]		
8	(a) (i)	hom	eostasis ;		[1]		
	(ii)	sma	Il intestine / ileum / duodenum ;		[1]		
	(iii)	secr	etes insulin ;		[1]		
	(iv)	diab	etes ;		[1]		
	(v)	from by d	ugh placenta ; n mother's blood ; iffusion ; ect reference to umbilical cord ;		[max 2]		

Page 6		;	Mark Scheme: Teachers' version IGCSE – October/November 2009				Syllabus 0654	Paper 02
(b)	(i) ger	notype	es of parents	<u> </u>	Aa		Aa	02
	gar	netes		A	and a	A) and (a)	
					gametes fr	om one pare	ent	
					A		a	
				A	1 AA can smell	2 A can s		
			gametes from other parent	a	3 Aa can smell	4 a cannot		
		all ga geno	and parent cor ametes correct otypes of offsp notypes of offs	t ; ring corre	ect ; rect (need not be ir	the boxes)	. ,	[4]
	(ii)	3 in 4	4 / 75 % / 0.7	5;				[1]
								[Total: 11]
9 (a)	(i)	(KE = 0.5	=)½ mv² ; 5 × 4000 × 0.5	5 × 0.5 = 5	;00 J ;			[2]
	(ii)		mentum =)m 00 × 0.5 = 20		S;			[2]
(b)	(i)	3 00	0 N ;					[1]
	(ii)	= 30	done = force 00 N × 2 = 60 w e.c.f. from k)00 J ;	е;			[2]
(c)			a = 1.6m ² ; = 40000 / 1.6	= 25000	N / m² ;			[2]

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(d) bloc	od los	ses heat through ears ;		
larg	jer su	rface area for radiation;	max 1	[1]
(e) (i)	num	ber of waves per second, etc. ;		[1]
(ii)	elep	hant , human and rabbit ;		[1]
(iii)	cat ;			[1]
				[Total: 13]