CO-ORDINATED	SCIENCES	0654/01
Paper 1 Multiple (Choice	
		May/June 200
		45 minutes
Additional Materials:	Multiple Choice Answer Sheet	
	Son clean eraser	

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A**, **B**, **C** and **D**.

Choose the one you consider correct and record your choice in soft pencil on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer. Any rough working should be done in this booklet. A copy of the Periodic Table is printed on page 20.

This document consists of **17** printed pages and **3** blank pages.



1 The diagram shows a bone from the human arm.



Use the key to identify the bone.

i	Has a distinct socket at one end Has no distinct socket at one end	go to ii go to iii
ii	Broad and flat in shape Rod-like in shape	A B
iii	Has a rounded structure that fits into a socket Has no rounded structure	C D

2 The diagram shows four cells.

Which cell is a plant cell?



3 The diagram shows two stages in the growth of a plant inside a black box.

Light enters the box through a hole in one side.

Which part is the effector?



day 1

day 5

4 Air is moved in and out of the lungs by changes in the volume of the chest cavity.

In which structure are there muscles that help to bring this about?

- A alveolus
- B bronchus
- C diaphragm
- D trachea
- 5 From which chamber of the human heart is blood pumped most strongly?



6 The graph shows how a person's body temperature changes with changing air temperature.



Which process provides the energy for maintaining the body temperature as shown in the graph?

- A breathing
- **B** digestion
- **C** excretion
- **D** respiration
- 7 A food contains reducing sugar, but no starch.

What colours will be obtained if samples of the food are tested with Benedict's reagent and with iodine solution?

	Benedict's test	iodine test
Α	blue	blue-black
В	blue	brown
С	red-orange	blue-black
D	red-orange	brown

- 8 Which bones form a joint at the shoulder?
 - A humerus and radius
 - B humerus and scapula
 - C ulna and radius
 - D ulna and scapula

9 The graphs show changes in the amount of sugar in the blood after a person has eaten a sugary meal.

Which graph shows changes in the amount of blood sugar of a person with untreated diabetes?



10 Where does fertilisation take place in a flowering plant?

- A anther
- **B** bud
- C ovule
- D stigma
- 11 An organism has 28 chromosomes in each body cell.

How many chromosomes would there be in a gamete of the same organism?

A 7 B 14	C 28	D 56
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Which shows this food relationship?



- 13 Which organisms increase the amount of nitrogen gas in the air?
 - A decomposing bacteria
 - **B** denitrifying bacteria
 - **C** nitrifying bacteria
 - D nitrogen fixing bacteria
- **14** The diagram shows a sack containing a mixture of three minerals.



Which element is not present in the mixture?

- A cobalt
- B copper
- **C** iron
- D tin

15 Heating a metal compound in a Bunsen flame turns the flame green.

Which metal ion is present in the compound?

- A calcium
- B copper
- C potassium
- D sodium
- 16 Which process produces molecules with longer chains?
 - A combustion of hydrocarbon
 - B cracking
 - C fractional distillation of crude oil
 - **D** polymerisation
- **17** The diagram shows part of a polymer molecule.



Which polymers can be represented by this diagram?

	cellulose	protein	starch
Α	\checkmark	\checkmark	\checkmark
в	1	\checkmark	×
С	1	×	\checkmark
D	x	\checkmark	\checkmark

18 The table shows the name and formula of four metal ores.

	name	formula
1	chalcopyrite	CuFeS ₂
2	ilmenite	FeTiO ₃
3	malachite	Cu ₂ CO ₃ (OH) ₂
4	wolframite	FeWO ₄

Which metal ores contain two different metals?

- A 1 and 3 only
- **B** 2 and 4 only
- **C** 1, 2 and 4 only
- **D** 2, 3 and 4 only
- **19** Compound **X** is an important industrial raw material.

Products obtained from the electrolysis of its concentrated aqueous solution include:

- a gas that bleaches damp litmus paper,
- a gas that ignites with a pop,
- an alkali.

What is compound **X**?

- **A** copper(II) sulphate
- **B** hydrochloric acid
- **C** sodium carbonate
- D sodium chloride

20 Small amounts of barium nitrate and sand are shaken with separate samples of water in two test-tubes. The test-tubes are left to stand.

Which two diagrams show the behaviour of the barium nitrate and sand?



21 A sample of tap water forms a scum, rather than a lather, with soap solution.

This shows that the tap water is ...X... and that it contains ...Y... ions.

What are X and Y?

	Х	Y
Α	hard	calcium
В	hard	sodium
С	soft	calcium
D	soft	sodium

22 Modern synthetic dyes have largely replaced natural plant dyes.

This is because, compared with natural plant dyes, synthetic dyes ...X... readily fade in sunlight and can be made in a ...Y... choice of colours.

Which words correctly complete gaps X and Y?

	Х	Y
Α	less	bigger
В	less	smaller
С	more	bigger
D	more	smaller

- **23** How may the compounds in chlorophyll be separated?
 - A chemotherapy
 - B chromatography
 - C distillation
 - D emulsification
- **24** A fuel used for cooking food is the hydrocarbon ...X... that burns in an ...Y... reaction.

Which words correctly complete gaps X and Y?

	Х	Y
Α	coke	endothermic
В	coke	exothermic
С	methane	endothermic
D	methane	exothermic

25 The table shows the results of adding three metals, X, Y and Z, to water and to dilute hydrochloric acid.

motal	reaction with	
metai	water	dilute hydrochloric acid
х	no reaction	no reaction
Y	violent reaction	explodes
Z	bubbles slowly	bubbles vigorously

What is the order of reactivity of the three metals?

	most reactive		least reactive
Α	Х	Y	Z
В	Y	х	Z
С	Y	Z	х
D	Z	Y	Х

26 An element X has a high melting point and its oxide, X_2O_3 , is coloured.

How are X and X_2O_3 described?

	Х	X_2O_3
Α	transition metal	acidic
в	transition metal	basic
С	non-metal	acidic
D	non-metal	basic

27 The positions of four elements are shown on the outline of part of the Periodic Table.

Which element would form an ion with two positive charges?



28 A measuring cylinder contains some water. When a stone is put in the water, the level rises.



- **A** 50 cm^3 **B** 70 cm^3 **C** 75 cm^3 **D** 125 cm^3
- **29** A child is standing on the platform of a station, watching the trains.



- **30** Which form of energy do we receive directly from the Sun?
 - A chemical
 - B light
 - **C** nuclear
 - D sound

31 The diagram shows a thick sheet of glass.

Which edge must it stand on to cause the greatest pressure?



32 Two plastic cups are placed one inside the other. Hot water is poured into the inner cup and a lid is put on top as shown.



Which statement is correct?

- A Heat loss by radiation is prevented by the small air gap.
- **B** No heat passes through the sides of either cup.
- **C** The bench is heated by convection from the bottom of the outer cup.
- **D** The lid is used to reduce heat loss by convection.
- 33 Which is the best description of the speed of a water wave?
 - A the distance between one wave crest and the next
 - **B** the distance between the crest of a wave and a trough
 - **C** the distance that a particle of water moves up and down in one second
 - **D** the distance that a wavefront moves along the surface in one second

34 A ray of light passes through a window.

Which path does it take?



35 Sounds are made by vibrating objects. A certain object vibrates but a person nearby cannot hear any sound.

Which statement might explain why nothing is heard?

- **A** The amplitude of the sound waves is too large.
- **B** The frequency of the vibration is too high.
- **C** The sound waves are transverse.
- **D** The speed of the sound waves is too high.
- **36** A student investigates which end of a magnetic compass needle is attracted to a bar magnet.

What does the investigation show?

- **A** Both ends of the compass needle are attracted by the north pole of the magnet.
- **B** Both ends of the compass needle are attracted by the south pole of the magnet.
- **C** One end of the compass needle is attracted by the north pole and the other end by the south pole.
- **D** The compass needle is not attracted by either end of the magnet.

37 Four lamps and four switches are connected to a power supply as shown in the circuit diagram.When all the switches are closed, all the lamps are lit.

When one of the switches is then opened, only one lamp goes out.

Which switch is opened?



38 An electric power tool is being used outdoors in a shower of rain.

What is the greatest hazard to the user?

- A The cable gets hot and causes burns.
- **B** The circuit-breaker cuts off the current.
- **C** The current passes through water and causes a shock.
- **D** The tool rusts.

- **39** A Geiger counter detects radiation from radioactive sources.
 - A radioactive source is inside a thick aluminium container as shown.



thick aluminium container

Which type of radiation from this source is being detected?

- A alpha-particles
- B beta-particles
- **C** gamma-rays
- D radio waves
- 40 A thermistor is a device whose resistance decreases as its temperature increases.

The table shows the voltage needed at different times during the day to cause a current of 0.02A in a particular thermistor.

time of day	09:00	12:00	15:00
voltage/V	12.0	6.0	4.0

Which statement describes how the temperature changed during the period 09:00 to 15:00?

- **A** The temperature decreased throughout this period.
- **B** The temperature increased throughout this period.
- **C** The temperature was greatest at 12:00.
- **D** The temperature was least at 12:00.

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					64 Copper 29	108 Ag Silver	197 Au Gold 79		157 Gd Gadolinium 64	Corrium Corrium
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			_		59 Co Cobalt	103 Rhodium 45	192 Ir Iridium		150 Samarium 62	Plutonium
		Hydrogen	-		56 Iron 26	101 Ru Ruthenium 44	190 OSmium 76		Promethium 61	Neptunium
					55 Mn Manganese 25	Tc Technetium 43	186 Re Rhenium 75		144 Neodymium 60	238 Uranium
					52 Chromium 24	96 Mo Molybdenum 42	184 V Tungsten 74		141 Pr Praseodymium 59	Protactinium
					51 Vanadium 23	93 Niobium 41	181 Ta Tantalum 73		140 Ce Cenium 58	232 Thorium
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