

**MARK SCHEME for the October/November 2013 series**

**0653 COMBINED SCIENCE**

**0653/23**

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page 2	Mark Scheme	Syllabus	Paper
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- 1 (a) A to cell membrane/membrane round vacuole ;  
 B to nucleus ;  
 C to cell wall/large vacuole ; [3]
- (b) water ;  
 mineral (salts)/named mineral ; [2]
- (c) (i) transport ;  
 of sugars/substances made in the leaves ; [2]
- (ii) roots have no sucrose/short of nutrients ;  
 no source of energy/cannot respire ; [2]
- [Total: 9]**
- 2 (a) (i) argon/Ar ; [1]
- (ii) calcium/lithium and oxygen/sulfur/fluorine ;  
 metal with non-metal ; [2]
- (b) (i) nucleus ; [1]
- (ii) 15 ;  
 same as number of electrons/3 shells = Period 3, 5 outer electrons means Group 5 so  
 must be phosphorus which has proton number 15 ; [2]
- [Total: 6]**
- 3 (a) change resistance (of circuit)/change current through resistor ; [1]
- (b) X – ammeters need to be in series in a circuit ; [1]
- (c) 13.3 ;  
 ohms ; [2]
- [Total: 4]**

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4 (a)

<b>nutrient tested for</b>	<b>reagent</b>	<b>result</b>	<b>conclusion</b>
<b>protein</b>	biuret	<b>purple</b>	contains protein
<b>starch</b>	iodine (solution)	blue black	<b>contains starch</b>
<b>fat</b>	ethanol (and water)	<b>milky white</b>	contains fat

one mark for any two correct ; ; ; [3]

(b) for growth  
for repair  
other use of protein ; ; [max 2]

(c) increases surface area ;  
idea of making it easier for enzymes to make contact ; [2]

(d) soil erosion  
flooding  
increases carbon dioxide in atmosphere  
extinction  
drought  
increase in global warming ; ; [max 2]

**[Total: 9]**

5 (a) (i) **Q** ;  
orange layer is rust ;  
formed when iron reacts with (dissolved) air and water ; [3]

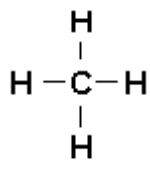
(ii) calcium/magnesium/zinc ;  
hydrogen ; [2]

(iii) (acid) temperature was lower/metal in **P** was less reactive/metal in **P** had smaller surface area ; [1]

(b) (i) different boiling points ; [1]

(ii) (vehicle) fuel ;  
burns easily/releases much energy when burnt ; [2]

(c) (i)



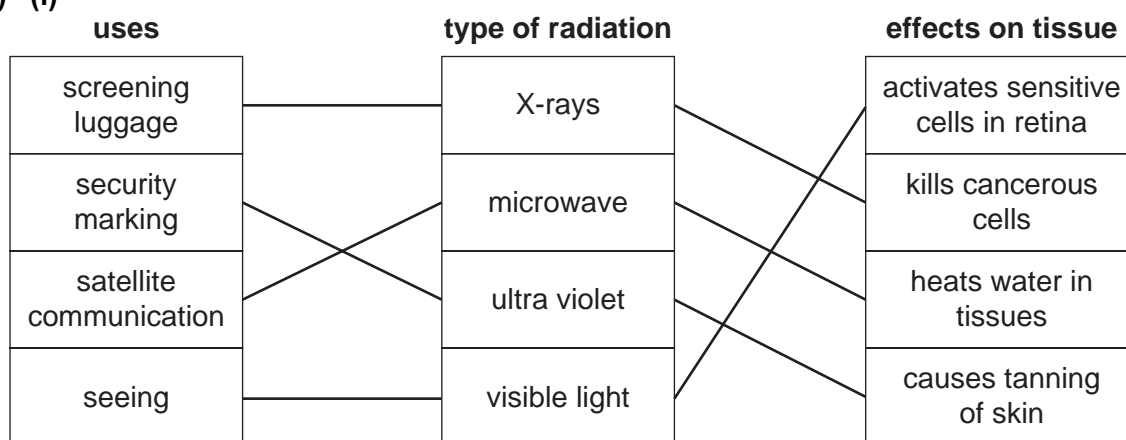
[1]

(ii) (methane) + oxygen → carbon dioxide + water ; ;  
(LHS and RHS)

[2]

**[Total: 12]**

6 (a) (i)



Each side, 3 correct 2mk, 1 correct 1mk

[4]

(b) sine wave ;  
amplitude correctly labelled ;  
wavelength correctly labelled ;

[3]

(c) (i) same horizontal level as nose ;  
same distance behind mirror that nose is from mirror ;

[2]

(ii) same size as object ;  
upright ;  
virtual ;

[3]

**[Total: 12]**

Page 5	Mark Scheme	Syllabus	Paper
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- 7 (a) A trachea ;  
B lung ; [2]
- (b) (i) movement of molecules ;  
from region of high concentration to low concentration/down a concentration gradient ; [2]
- (ii) plasma ; [1]
- (iii) more energy used/more muscle contraction ;  
more carbon dioxide produced by cells ;  
in respiration/by oxidation of glucose ; [max 2]
- (iv) increases/faster ;  
greater diffusion gradient (from blood to alveolus) ; [2]
- [Total: 9]**
- 8 (a) (i) gas into limewater ;  
goes cloudy ; [2]
- (ii) salt ; [1]
- (iii) 9 ;  
3 ; [2]
- (b) (i) the higher the pH , the less acidic/the lower the acid concentration ;  
lowering acid concentration decreases the rate ; [2]  
(accept correct collision theory explanation)
- (ii) temperature  
surface area of calcium carbonate  
degree of agitation of the mixture ; ; [max 2]
- [Total: 9]**

Page 6	Mark Scheme	Syllabus	Paper
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- 9 (a) (i) 80 m ; [1]
- (ii) not moving ; [1]
- (iii) unbalanced because speed is changing ; [1]
- (b) geothermal/hydroelectricity/waves/wind/biomass ; [1]
- (c) (i) kinetic energy ; [1]
- (ii) gravitational potential energy ; [1]
- (d) density = mass/volume ;  
=  $45/36 = 1.25 \text{ g/cm}^3$  ; [2]
- (e) particles far apart ;  
irregular arrangement ; [2]

**[Total: 10]**