## MARK SCHEME for the October/November 2012 series

## **0653 COMBINED SCIENCE**

0653/31

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



	Page 2				Syllabus		
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1	a c a p me uni	omple article asure t of po	nt given ete loop of conductors e with an negative electrical charge es potential difference ower ect for 2 marks, 2 or 3 correct for 1 mark)	word require circuit ; electron ; voltmeter ; watt ;	ed	[max 2]	
	(b) (i)	. –	es out) mplete circuit ;			[1]	
	(ii)	so th	nat they can be individually turned on and nat they all get the full mains voltage ; nat if one fails the rest still operate ;	d off ;		[max 2]	
	(iii)	= 1/	.= 1/ <b>R</b> <sub>1</sub> + 1/ <b>R</b> <sub>2</sub> ; 1.2 + 1/1.2; 0.6Ω;			[3]	
			,			[Total: 8]	
2	(a) (i)		, <b>F</b> ; (all required)			[2]	
	(ii)	stard	ch/cellulose/sugar/any other correct ;			[1]	
	ani use for	mals) e carb respir	dead (plant or animal) material/was ; oon-containing substances/sugar ; ration ; arbon dioxide to the air ;	te products	(from plants or		
			,			[max 2]	
	(c) (i)	(the	that the graph shows a maximum ; maximum occurs) at 480 ± 20 Hz ; of steeper decrease than increase ;			[2]	
	(ii)		vl out of soil when they hear vibrations (o re moles cannot catch them ;	f 500 Hz) ;		[2]	
						[Total: 9]	
_		_					
3	(a) (i)	>7 u	up to 14 ; down to 1 ;			[2]	
	(ii)	mete	er is more accurate/precise/reference to	quantitative	<b>;</b>	[1]	
	(iii)		(acidified) silver nitrate/ethanoate (solut e precipitate/solid indicates hydrochloric		le (ions) ;	[2]	

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Page 3			Paper					
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(b)	(i) (ii)	(no r	nesium + copper oxide — magnesium oxide + reaction) eacts with copper oxide ;	+ copper ;				
		<b>G</b> m	ore reactive than copper ; opper cannot remove oxygen from oxide of <b>G</b> /owt	te ;	[max 2]			
					[Total: 8]			
(a)	(wc	ork do	60 kg so weight is 600 N ; ne =) force × distance/(P.E. gained =) gravitationa 1.3 = 780 J ;	al field strength × h	eight ; [3]			
(b)	780	)(J);			[1]			
(c)	(c) power = work/time ; 780/0.5 = 1560 W ;							
					[Total: 6]			
(a)	(i)		$_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$ ;; hand side and right hand side)		[2]			
	(ii)	volui	oon dioxide would not be absorbed ; me of carbon dioxide produced = volume of oxyger o change in volume ;	n used ;	[max 2]			
(b)	(i)	as a seec	control/to check that movement was caused by gods ;	erminating/living	[1]			
	(ii)	10°C	eased rate of respiration with increased temperatur C rise doubles rate ; of data linking distance moved and rate of respirati	-	ion ; [max 2]			
(iii)			novement ;		[0]			
		enzy	vmes do not work at high temperatures/enzymes d	ienatured ;	[2]			

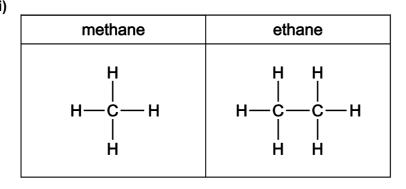
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Page 4		4	Mark Scheme Syllabu			Syllabus	;	Paper			
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6	<ul> <li>(a) reference to high rate/fast reaction ; powder has high surface area ; high surface area (of solids) increases rate/ref. to contend to the surface area (of solids) increases rate/ref.</li> </ul>				/ref. to co	ollision fr	requency ;		[max 2]		
	(b) (i)	<ul> <li>(b) (i) electrons transferred from aluminium (atoms) to oxygen (atoms); reference to filled outer shells; reference to formation of positive and negative ions; correct detail i.e. aluminium ions positive and oxide ions negative; strong (force of) attraction between ions;</li> </ul>							[max 4]		
	(ii)	bala refer	rence to	quires sa the oxyg	en imbala	ance ;			n both sides es equation ;	;	[max 2]
	<ul> <li>(c) (components in) firework mixture must burn/require oxygen to burn/need to be oxidised;</li> <li>potassium perchlorate produces oxygen (when heated);</li> <li>idea that oxygen needs to be produced in situ/air cannot easily get into firework mixture;</li> </ul>					b	[max 2]				
											[Total: 10]
7	wa	veleng	e labelle gth labe limensio		th ;						[3]
	(b) (i)	<b>A</b> is	louder t	han <b>B</b> ;							[1]
			as highe								[1]
	(iii)	spee vacu solid liquid gas (all c	ed of sou m/s uum d correct fe	und, there 0; 5000; 1500; 330; or 2 mark		correct	for 1 mark)	)			[max 2]
	(iv)	regio <i>raret</i>	faction	h pressu		. , .	rticles/pai icles/parti		loser ; pre spaced ;		[2]

	Page 5		Mark Scheme	Syllabus	Paper	
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	(c)	· · · ·	on ; adiation can travel through vacuum/conduction and nedium ;	convection	[2]	
	(d)	•	ntinued as series of straight lines ; approx correct ;		[2]	
					[Total: 13]	
8	(a)	B se C	carries, sperm/semen ; produces fluid, for sperm to swim in/containing suga minal fluid ; carries, sperm/semen and urine ; pel to testis ;	r/secretes	[3] [1]	
		more n have a	ed in larger quantities ; nobile ; tail/pointed head/streamlined ;		[max 2]	
	(c)	referer	estroys/attacks white blood cells ; ce to (T) lymphocytes/T cells ; s ability to, destroy viruses/fight infection ;		[max 2]	

[Total: 8]

9 (a) (i)

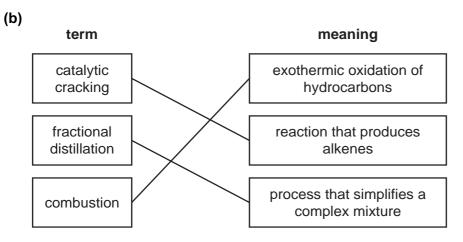


[2]

(ii) bottled gas ; heating ; cooking ;

[max 1]

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(all correct for 2 marks, 1 correct for 1 mark)

[2]

- (c) (i) decane reacted/decomposed/cracked; products include alkenes/ethene/unsaturation; alkenes react with bromine/decolorise bromine;
   [3]
  - (ii) makes catalyst more efficient/work better/increases reaction rate; [1]

[Total: 9]