CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2012 series

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



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1 (a) (i) haploid;

zygote;

dissimilar; [3]

(ii) fertilisation; [1]

(b) (i) A - anther/stamen;

B – stigma ; [2]

(ii) A;

D;

(c) (i)

tube	conditions			
С	water	oxygen	no light	
D	no water	oxygen	no light	
E	water	no oxygen	no light	

(all three tubes correct for 2 marks, two tubes correct for 1 mark)

[2]

(ii) (lettuce) seeds need oxygen (for germination); (lettuce) seeds need water/moisture (for germination); (lettuce) seeds do not need light (for germination); (max 2 marks if germination **not** mentioned)

[3]

[Total: 13]

2 (a) (i) nitrogen 78 (%);

oxygen 21(%); [2]

(ii) nitrogen/an element is in the Periodic Table/nitric oxide/a compound is not; nitrogen/an element only contains one type of atom/nitric oxide/a compound contains more than one type of atom/element; nitrogen/an element cannot be broken down into simpler substances/nitric oxide/a compound can;

the atoms in nitric oxide / a compound are bonded together; [max 2]

(iii) it represents one molecule of nitric oxide;

reference to the bonding of atoms; reference to the 1:1 ratio of N:O;

[max 2]

(iv) oxidation; [1]

(b) (i) ionic/electrovalent;

bonding is between metal and non-metal;

[2]

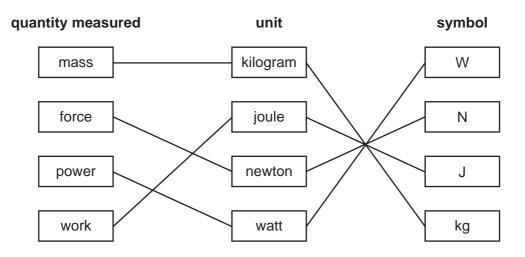
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(ii) products of combustion of magnesium reacted with water; to form an alkaline solution/an alkali;

[max 1]

[Total: 10]

3 (a)



[2]

- (b) A constant speed/velocity;
 - **B** constant acceleration;

[2]

(c) (distance covered =) speed × time;

 $20 \times 90 = 1800 (m)$;

[2]

(d) (i) (resistance =) voltage/current;

$$= 12/2 = 6 (\Omega)$$
;

[2]

[2]

- (ii) (R =) R1 + R2;
 - $6 + 6 = 12 (\Omega)$;

[Total: 10]

4 (a) carnivore and consumer;

[1]

(b) (i) any number above 20 000;

[1]

(ii) longitudinal;

- [1]
- (c) (i) idea of reflected randomly/scattered from rough surface/regularly from smooth;
- [1]
- (ii) idea that it receives ultrasound back from a rough surface (but not from smooth);
- [1]

	Page 4			Mark Scheme	Syllabus	Paper
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	(d)	flood	e car ling ;	bon dioxide in air/increased greenhouse effect/glo	bal warming ;	[max 3] [Total: 8]
5	(a)	filtrati remo AND evap	(harr //OR tion ; oves //OR oorat	mful) microorganisms ; ; ; solids ;		[max 4]
	(b)		red ; dye (giving only one spot matches red in P /owtte ;		[2]
		(ii)	c .			[1]
		(")	Ο,			
						[Total: 7]
6	(a)	heat kinet				[2]
	(b)	((as) (mor	er turns to a gas/(water) vapour; particles/molecules get further apart; re) energetic particles escape; is needed/used to cause evaporation;		[max 2]
		` ((mor	is needed/used to cause evaporation; re) energetic particles escape; aining (particles) have less (thermal) energy;		[max 1]
	(c)	<pre>in solid: particles in regular arrangement; particles all touching; in liquid: particles arranged irregularly; particles touching;</pre>			[max 2]	
	(d)			cy =) useful energy out ÷ energy in ; ow little or how much energy is wasted by machine	device ;	[max 1]

[Total: 8]

(a)	(i)	A – incisor/canine ; B – molar/premolar ;				[2]
	(ii)	crush/grind; break into small piece increase surface area idea of better access	of food;	ier to digest ;		[max 2]
(b)		part	ingestion	digestion	absorption	
		mouth	✓	✓		
		stomach		✓		
		small intestine		✓	✓	
	(1 r	mark per correct row) ;;	;			[3] [Total: 7]
(a)	(i)	ductile ; (electrical) conductor	;			[2]
	(ii)	mixture of metals; alloy is less malleable	/hard <u>er</u> /strong	<u>er</u> /low <u>er</u> melting լ	point ;	[2]
	(iii)	copper sulfide + oxyg	en ——► coppe	er + sulfur dioxide	;	[1]
(b)	(i)	copper chloride solution	on/the conducti	ng solution ;		[1]
	(ii)	chlorine; bubbles/gas given of	f;			
		copper; reference to copper co	oloured/brown/	pink layer/solid ;		[4]
						[Total: 10]

Mark Scheme
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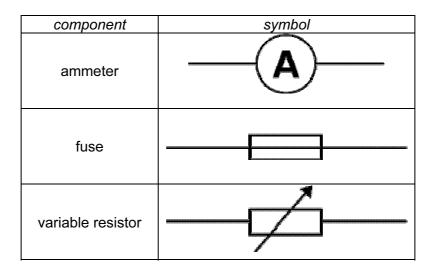
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8

Syllabus 0653 Paper 23

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9 (a)



[3]

(b) (i) 3;

(ii) correct symbol in parallel with bulb; [1]

(c) (i) angle of incidence and angle of reflection; [1]

(ii) 45°;

[Total: 7]