MARK SCHEME for the October/November 2011 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/63

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

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	Page 2		Mark Scheme: Teachers' version	Syllabus	Paper		
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1	(a) (i)	ехра	nata/pores ; anding gas , oxygen/CO ₂ ;		[max 2]		
	(ii)		between 42 and 45 ; between 20 and 24 ;		[2]		
	(iii)		no. of squares for C e.g. 42 multiplied by 100 ; no. of squares for P e.g. 20 multiplied by 100 ;		[2]		
	(iv)	more less more	ner ; rect sun ; e wind movement ; humid ; e water loss ; e wilting ;		[max 2]		
	• •	bundle <u>əm</u> ;	es indicated by shading ;		[2] [Total: 10]		
2	(a) (i)	gree to ye	n ; ellow/orange ;		[2]		
	(ii)	carb	onic acid ; (allow H ₂ CO ₃)		[1]		
	(b) (i)	turns	s white/white precipitate/milky/cloudy/owtte;		[1]		
	(ii)	white	e/milkiness disappears/owtte (reject dissolves/rea	cts) ;	[1]		
	(iii)	(g) =	= aqueous/dissolved ; gas/gaseous ; solid ;		[3]		
	(iv)	prec	ipitate ;		[1]		
	(c) B a	c) B and C ;					
					[Total: 10]		

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3	(a) () (i) 1.9 2.3		A; / (± 0.1);		[2]		
	(i	ii)		2.5 = 0.8 (e.c.f., accept answers with more decimal 1.9 = 1.2 ;	places) ;	[2]		
	(b) ((i)		[3]				
	(i	ii)		e extended to show five wires ; ut 0.5 ohms (value from candidate's graph) ;		[2]		
	• •	 (c) repeat (the experiment (using 1 wire – with different voltages and average results)); 						
						[Total: 10]		
4	(a) ((i)		C rate = 0.77/min ; C rate = 0.50/min ;		[2]		
	(b) ((i)		ect plotting ; eptable smooth curve drawn ;		[2]		
	(i	ii)	50°C	С;		[1]		
	(ii	ii)	canr	not tell exactly the rate either side of 50°C/owtte ;		[1]		
	(c) ((i)	(rate	e speeds up due to) particles moving faster/more co	Ilisions ;	[1]		
	(i	ii)	prote	ein denatures (due to high temperatures) ;		[1]		
	(d) t t			to check if acid is needed for the reaction ; to see if pepsin is needed/see if acid could do reac	tion ;	[2]		
						[Total: 10]		
5	(a) ((i)	wate	er, ethanol, propanone or any suitable named organ	ic solvent ;	[1]		
	(i	ii)	horiz	zontal line drawn below the start line ;		[1]		
	(ii	ii)	to pr	revent paper drying out/solvent evaporating/owtte ;		[1]		
	(iv	v)	any	reasonable length of time, e.g. between 30 and 180	minutes ;	[1]		

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	(b)	(i) (ii)	one o have	are mixtures/impure ; contains two dyes the other three ; one common dye ; is pure one a mixture/only 3 pure ;	[any 1]	
			one	contains three dyes the other one ; no common dye ;	[any 1]	
	(c)		ned ao ned al	cid ; Ikali (either order) ;	[1] [1]	
	cut spot from paper/use of spot ; add acid or alkali to spot ;					
		100	K TOP C	olour change ;	[max 2]	
					[Total: 10]	
e	(0)	50		rk loballad V :		
6	 (a) 58 cm mark labelled Y ; 51 cm mark labelled Z ; 					
	(b)	line	es YO	and ZO drawn (e.c.f.) ; (ruler straight)	[1]	
	(c)	(i)	66 m	m (or as candidate's diagram) ;	[1]	
		(ii)	63 m	m (or as candidate's diagram) ;	[1]	
		(iii)	87 m	m (or as candidate's diagram) all ± 1 mm ;	[1]	
	(d)	(i)	87/60	6 = 1.3 (e.c.f) ;	[1]	
	()	.,		3 = 1.4 (e.c.f);	[1]	
	(e)	(i)	•	ow) because the fish is deeper/further away than he sees it/light is y from the normal as it leaves the surface/owtte ;	s bent [1]	
		(ii)	his a owtte	im must be deeper than in fresh water, because the light is bent r e ;	more/ [1]	
					[Total: 10]	
					- •	