9*CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2013 series

0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



	Page 2		Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2013	0653	22
1	(a) (i)	hydr	ogen;		[1]
	(ii)	rate	increases (down the group);		[1]
	(b) (i)	melt	ing point increases (down the group) ;		[1]
	(ii)	bron	ure becomes orange; nine is produced/is orange; rine is more reactive than bromine/chlorine displac	ces bromine ;	[3]
	(c) (i)	reac does	sphorus is (very) reactive ; ts with substances in the air ; s not react with water ; er forms a barrier ;		[max 2]
	(ii)	phos (con (that		[max 2] [Total: 10]	
2	(a) (i)	friction	on ;		[1]
	(ii)	new	tons;		[1]
	(iii)		itational potential to kinetic ; mal/sound ;		[2]
	(iv)		ed = distance/time; 2/3 = 0.4 (m/s);		[2]
	(b) ford	ce ; cance	•		[2]
	(c) (i)		w 20 Hz ; an lower threshold is about 20 Hz ;		[2]
	(ii)	num	ber of vibrations per second ;		[1]
					[Total: 11]

		IGCSE – May/June 2013	0653	22
the	y have	e warmth and water/moisture ;		[3]
) (i)				[2]
(ii)	whe	re insects can reach them/attracts more insects;		[max 2]
				[Total: 7]
) (i)				[2]
(ii)	comi	munication ;		[1]
				[2]
				[Total: 5]
) (i)	com elem	pound contains different atoms that are bonded ; nents shown in Periodic Table/compounds are not s	hown ;	[max 2]
(ii)	natu	ral gas ;		[1]
(iii)	coal	/peat ;		[1]
(iv)				[2]
) (i)	mag	nesium oxide ;		[1]
(ii)			ι;	[2]
				[Total: 9]
	the light (ii) (ii) (iii) (iii) (iv) (ii)	they have light is not light in the sense. (ii) geot sense where for possible in the sense where for possible in the sense where light is not light in the sense where light is not light in the sense where light is not light in the sense where light in the	 B and C; they have warmth and water/moisture; light is not needed; (i) geotropism; sensitivity; (ii) flowers held up; where insects can reach them/attracts more insects; for pollination; (i) thermal; thermal and conduction; (ii) communication; (iii) communication; (ii) regular arrangement; all touching; (ii) H and C/elements contain only one type of atom; compound contains different atoms that are bonded; elements shown in Periodic Table/compounds are not s compound has different properties from either element; (ii) natural gas; (iii) coal/peat; (iv) carbon dioxide; water; (iv) magnesium oxide; 	B and C; they have warmth and water/moisture; light is not needed; (i) geotropism; sensitivity; (ii) flowers held up; where insects can reach them/attracts more insects; for pollination; (ii) thermal; thermal and conduction; (iii) communication; regular arrangement; all touching; (i) H and C/elements contain only one type of atom; compound contains different atoms that are bonded; elements shown in Periodic Table/compounds are not shown; compound has different properties from either element; (ii) natural gas; (iii) coal/peat; (iv) carbon dioxide; water; (i) magnesium oxide; (ii) magnesium: atoms lose electrons/become a positive ion;

Mark Scheme

Syllabus

Paper

Page 3

	Page 4		Mark Scheme	Syllabus	Paper	
			IGCSE – May/June 2013	0653	22	
6	(a) (i)	chlo	rophyll ;		[1]	
	(ii)	carb wate	on dioxide ; er ;		[2]	
	(iii)	oxyg	gen ;		[1]	
	(b) an (fro	m) or				
		anima	[max 2]			
	rep	c) growth ; repair ; for making, cell membranes/cytoplasm ;				
			c substance ;	[max 2]		
	(d) (i)	last	three boxes ticked ;		[1]	
	(ii)	from (so) by re	e heat lost in cold environment; skin/by radiation/by conduction; more heat needs to be produced within the body/ir espiration;	n cells ;		
		to in	g, food/glucose/carbohydrates (as fuel); crease fat deposits under the skin; leat insulation;		[max 2]	
			,		[Total: 11]	
7	(a) (i)	lamp				
		cell ; swite	,		[3]	
	(ii)	corre	ect series circuit and all symbols correct ;		[1]	
	(b) (i)	geot	thermal/wave/tidal/hydroelectric/wind/biomass;		[1]	
	(ii)	coal	/oil/gas/peat/nuclear;		[1]	
	(iii)		duction requires particles/a medium; radiation can pass through a vacuum;		[max 1]	
		(c) angle of reflection ; 45°;				
					[Total: 9]	

	Page 5		Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2013	0653	22
8	(a)	P ; R ; Q, R	ł;		[3]
	(b) (i)	any	value from 8 to 14/8 – 14 ;		[1]
	(ii)	рН с	of '7' on the screen/owtte;		[1]
	(iii) (B		the least volume (to neutralise the alkali);		[1]
	(iv)	reac	tion was exothermic/heat given off;		[1]
	(v)	\rightarrow	salt; + water;		[2]
					[Total: 9]
9	(a) (i)	cont sma	ucleus ; ains haemoglobin ; ller ; dent in the middle ;		[max 2]
	(ii)		sports oxygen ; Iungs to, tissues/cells ;		[2]
		protection against disease/destroys invading microorganisms ; phagocytosis ;			
	sm	soluble ; small intestine ; adrenaline ;			
					[3] [Total: 9]