## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

## MARK SCHEME for the May/June 2006 question paper

## **0653 COMBINED SCIENCE**

0653/03

Paper 3, maximum raw mark 80

These mark schemes are published as an aid to teachers and students, to indicate the requirements of the examination. They show the basis on which Examiners were initially instructed to award marks. They do not indicate the details of the discussions that took place at an Examiners' meeting before marking began. Any substantial changes to the mark scheme that arose from these discussions will be recorded in the published *Report on the Examination*.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the Report on the Examination.

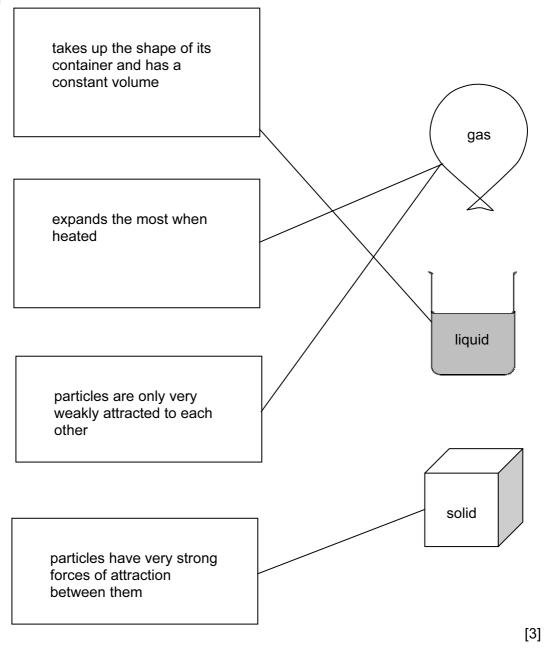
The minimum marks in these components needed for various grades were previously published with these mark schemes, but are now instead included in the Report on the Examination for this session.

• CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2006 question papers for most IGCSE and GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Page 1	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2006	0653	03

1 (a)



(b) particles moving randomly; particles collide with the walls of container; collide more often; so greater force exerted on walls of container; increase in pressure;

[Total: 6]

[3]

9		- J	
	IGCSE – May/June 2006	0653	03
` ,	ety of a gene ; only has an effect when dominant one not present ;		[2]
(b) (i)	aa ;		[1]
	parents are Aa and Aa ; gametes from each parent are A and a ; offspring are AA, Aa (twice) and aa ;		[3]
mea	ds containing) proteins ; t / fish / cheese / other e.g. of high protein food ; eins are made up of amino acids ;		[2 max]
			[Total: 8]

**Syllabus** 

Paper

Mark Scheme

Page 2

Page 3		Mark Scheme	Syllabus	Paper
		IGCSE – May/June 2006	0653	03
3	(a) (Cl) (Br) (I)	gas liquid solid;		[1]
	. , . ,	four shared pairs; symbols correctly shown for each atom;		[2]
	(ii)	$4Cl_2 + CH_4 \rightarrow CCl_4 + 4HCl;$		[1]
	` '	(fluorine) reactivity decreases down Group 7 / owtte;		[1]
	. , . ,	(nucleus of) $Cl - 37$ contains more neutrons than $Cl - 38$ 2 more;	5;	[2]
		(12 x 1) + (4 x 35.5); = 154;		[2]

[Total: 9]

	IGCSE – May/June 2006	0653	03
4 (a) (i)	reference to emission; reference to products;		[2]
(ii)	relatively short half life but not too short;		[1]
(iii)	3 half lives; so 0.2 g;		[2]
(b) (i)	high voltage means low current; this reduces energy losses;		[2]
(ii)	resistance = voltage/current; = 22 ohms		[2]
		Γ	Total: 9]

**Syllabus** 

Paper

Mark Scheme

Page 4

Page 5	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2006	0653	03
three	rrect two marks correct one mark or two correct no marks		[2]
v b	ef to water molecules; vater passes from beaker through ppm; because more water outside than inside / correct ref to starch (molecules) cannot pass through the membrane	-	[3 max]
	add iodine (solution) ; orange / brown / yellow ;		[2]

[Total: 9]

Page 6		6	Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2006	IGCSE – May/June 2006 0653	
6	(a)	car	bon dioxide / carbon monoxide / carbon / soot / water; (any	two)	[1]
	(b)	(i)	( <b>B</b> is methane) methane molecules have five atoms (bonded) / is CH <sub>4</sub> ;		[1]
		(ii)	(addition) polymerisation; many small molecules / monomers join to form a long chai	n;	[2]
	(c)	(i)	heated / vaporised / boiled; passed over catalyst;		[2]
		(ii)	suggests that only single bonds between carbon atoms / s if double bonds present bromine would have been decolor		[2]
					[Total: 8]

Page 7	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2006	0653	03
		•	

7 (a) leaves / plants, stop rain hitting the ground (hard);

roots hold soil in place;

terracing stops water running down slopes;

[2 max]

(b) colourless / green / small / no petals / dangling anthers / dangling stigmas; [1]

(c) (i) by diffusion;

through wall of small intestine;

ref. to villi;

[2 max]

(ii) pancreas;

secretes insulin;

causes, cells / liver, to take glucose from the blood;

[3]

(iii) as level moves away from norm;

process initiated to bring it back;

take these points from a specific description

[2]

[Total: 10]

Page 8	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2006	0653	03

(a) (i) element made of only one type of atom and compound contains different atoms bonded together;
element cannot be simplified and a compound can be broken into its elements / is made from different elements;

(ii) Fe<sup>3+</sup>; working refers to charge balance; (reject vague criss cross answers)

[2]

(b) a layer of zinc covers the steel / provides a barrier;
prevents reaction between steel and oxygen and water;
(allow correct references to sacrificial protection)

[2]

(c) (i) H<sup>+</sup>;

[1]

(ii) no more gas evolved;

[1]

(iii) grey crystals appear / magnesium reacts and dissolves; a metal displacement reaction occurs / or equation; because magnesium more reactive than zinc;

[3]

[Total: 10]

Page 9		)	Mark Scheme	Syllabus	Paper
			IGCSE – May/June 2006	0653	03
9	(a)	(i) a	acceleration;		[1]
		(ii)	constant speed;		[1]
	(b)		under curve = ; (or other suitable) + 25 = 175 m;		[2]
	(c)	equa	I and opposite/ balanced		[1]
	(d)		orce = mass x acceleration; = 120 N;		[2]
			power = work/time; = 600 W;		[2]
	(e)	lowe	no mark st CoG;		ro1
		base	wider than <b>P</b> ;		[2]
				רן	otal: 11]