UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the June 2005 question paper

0653 COMBINED SCIENCE

0653/02 Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which Examiners were initially instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began. Any substantial changes to the mark scheme that arose from these discussions will be recorded in the published *Report on the Examination*.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the Report on the Examination.

CIE will not enter into discussion or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the June 2005 question papers for most IGCSE and GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

Grade thresholds for Syllabus 0653 (Combined Science) in the June 2005 examination.

	maximum	minimum mark required for grade:				
	mark available	А	С	E	F	
Component 2	80	N/A	42	27	20	

The threshold (minimum mark) for D is set halfway between those for Grades C and E. The threshold (minimum mark) for G is set as many marks below the F threshold as the E threshold is above it.

IGCSE

MARK SCHEME

MAXIMUM MARK: 80

SYLLABUS/COMPONENT: 0653/02

COMBINED SCIENCE Paper 2 (Core Theory)

Page 1	Mark Scheme	Syllabus	Paper
	IGCSE – JUNE 2005	0653	2

1	(a)	D; A;			
		C;		[3]	
	(b)	(i)	xylem/vascular bundles;	[1]	
		(ii)	osmosis (from within the cell); as vapour/evaporation; by diffusion; through stomata;	max[2]	
2	(a)	•	centa/umbilical cord/umbilical vein;		
		blo am	od; niotic;	[3]	
	(b)	(i)	virus may get into baby's blood/cross the placenta; (reject any suggestion that baby shares mother's blood)	[1]	
		(ii)	avoid/reduce sexual contact/use of condom;	[1]	
	(c)	needed for bone formation; needed for formation of teeth; for the developing fetus/for the baby; to stop her teeth losing calcium;			
3	(a)	(i)	argon; noble/inert gas/reference to atoms being stable/having full outer shells;	[2]	
		(ii)	magnesium and oxygen/hydrogen;	[1]	
		(iii)	magnesium and argon;	[1]	
		(iv)	H/hydrogen; nucleon number and proton number both 1/other reasonable;	[2]	
	(b)	sar	ne number of each type of atom/element on both sides;	[1]	
	(c)	(i)	acid concentration;	[1]	
		(ii)	(3) has the highest rate of reaction; high surface area and high temperature both increase rate;	[2]	
4	(a)	(i)	<= 20 Hz;	[1]	
		(ii)	molecules vibrate; vibration passes from one molecule to another; description of compression and rarefaction;	max [2]	
	(b)	(i)	speed = distance ÷ time/suitable symbolic version; 21 ÷ 7/3 (cm/s);	[2]	

	F	Page	2	Mark Scheme	Syllabus	Paper	7
				IGCSE – JUNE 2005	0653	2	
		(ii)	0.02 x	force x distance/suitable symbolic version; (21 ÷ 100);			
			0.0042	2 (J);			[3]
	(c)		•	emitter of radiation;			
		air	•	ir; conductor (of heat); er suitable alternatives)			[2]
5	(a)	(i)	mucus	peat) and move mucus; s contains bacteria/dirt; ds/away from lungs;		ma	ax [2]
			apwar	as/away irom langs,		1110	1X [*]
		(ii)	bacter	op working; ia get into lungs/remain in lungs; ia cause damage/disease;		ma	ax [2]
	(b)	(i)	650;				[1]
		(ii)	as sulp	phur dioxide increases, deaths increase;			[1]
		(iii)		takes time to develop/other reference to time del reference to some sulphur dioxide still in the air)	•		[1]
6	(a)	(i)	potass	sium + chlorine → potassium chloride;			[1]
		(ii)		s/burning; ermic means heat is given out;			[2]
	(b)	(i)	neutra	lisation;			[1]
		(ii)	-	chloric (acid); sium hydroxide;			[2]
		(iii)		named indicator; nce to the corresponding neutral colour;			
			use a	pH meter; pH = 7;			[2]
		(iv)	evapo	heat the solution; rate/boil off the water; le additional practical detail;			[2]
7	(a)	(i)	arrow	labelled B pointing downwards;			[1]
		(ii)	forces	are equal/balanced;			[1]

[1]

(iii) gravity/weight;

Page 3	Mark Scheme	Syllabus	Paper
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	(b)	density = mass \div volume/suitable symbolic version; 25 \div 30 = 0.83(3) (g/cm ³);					
	(c)	par	diagram Y ; particles touching/close packed/very close together; in a regular arrangement;				
8	(a)	(i)	enzyme/carbohydrase/amylase;	[1]			
		(ii)	small intestine;	[1]			
		(iii)	move food along/mix with enzymes;	[1]			
	(b)	(i)	use of Benedict's solution and heat/warm ; colour change to orange/red;	[2]			
		(ii)	oxygen; carbon dioxide; water; (all three = 2 marks; two correct = 1 mark)	[2]			
9	(a)	coa	al;	[1]			
	(b)	(i)	(A) oxygen is reacting/joining with the wood/oxygen is needed for burning;	[1]			
		(ii)	carbon dioxide/carbon monoxide/water vapour;	[1]			
		(iii)	(thermal) decomposition; wood is made of large/complex molecules and simpler ones are being formed/owtte;	[2]			
	(c)	(i)	appearance of orange/brown/shiny substance/substance;	[1]			
		(ii)	carbon + copper oxide → copper + carbon dioxide/monoxide;	[1]			
10	(a)		istance = voltage ÷ current/suitable symbolic version; 0.5/10 (ohms);	[2]			
	(b)	description of convection; hot water rises; hot water is less dense;					
	(c)	spe	nsverse; eed; raction; io;	[4]			