UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2008 question paper

0652 PHYSICAL SCIENCE

0652/03

Paper 3 (Extended), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Page 2	2	Mark Scheme	Syllabus		Paper
			IGCSE – October/November 2008	0652		03
1	(a) (i)	use (of weight = mass x <i>g</i> ; O N ;		1 1	2
	(ii)	2.0 N	N OR same as (i) ;		1	1
	(b) arro	ow ve	rtically upwards ; (allow without label if clear)		1	1
	(c) mai	rked c	clearly between 5.0 & 5.5 N ;		1	1
	(d) (i)	1.9 ±	Ŀ 0.1N ;		1	1
	(ii)		of force = mass x acceleration ; 5 m/s ² ;		1 1	2
						[Total: 8]
2	(a) (i)	coati	ing with zinc ;		1	1
	(ii)	whe	is more reactive than iron; n both exposed to water and oxygen zinc corrodes/lecting the iron/sacrificial corrosion;	reacts ;	1 1 1	3
	(iii)	pain	ting;		1	1
	(iv)	OR f	aint/oil/grease etc: no, if scratched the iron rusts/ for stainless steel: yes, because protection is throug alloy not just on the surface	hout the	1	1
			m has an oxide layer ; events contact between the metal and oxygen/air/w	ater ;	1 1	2
	(c) (i)	mak	es it stronger ;		1	1
	(ii)	(ns of second metal get between aluminium metals in of the two metals are of a different size; ing it more difficult for layers of atoms to slide;	n lattice/atoms	1	2

[Total: 11]

	Page 3			Mark Scheme Sylla		Paper
				IGCSE – October/November 2008	0652	03
3		the liquid moves up the capillary tube ; because it expands ;				2
	(b)	(i)	iron,	copper, constantan ANY TWO	1+	1 2
	((ii)		perature = 100 × 4.8/7.2 ; 67°C ;	1 1	2
	(can r	c acting OR measure higher temperatures OR be remote ;	1	
			meta	hermal capacity or can follow changing temps OR als used have Higher melting points than glass OR is can be as long as required;	1	2
						[Total: 8]
4		2,8,8 2,8,8 2,5 ;	3;	1 1 1	3	
				of electrons in outer shell ; Group number	1	2
	(c)	(i)	CaI ₂	;	1	1
	((ii)	black	(accept dark grey/blue);	1	1
	(d)			ng point increases ; increase in proton number/down Group ;	1 1	2
	(argo	m is less dense than air so will float/carry balloon up n and krypton are more dense than air so will not floa n only slightly less dense than air, will not give enougl	at/will sink; 1	
				not make balloon rise ;	1	3
					[Total: 12]	

	Page 4			Mark Scheme Syllal	bus	Paper
				IGCSE – October/November 2008 065	52	03
5	(a)	refr wav	actior velenç	efracted on entering shallow water ; n correct ; gth in deep water constant AND in shallow water ; wavefronts drawn max. 2, 2 drawn max 1)	1 1 1	3
	(b)	(i)	not r	circles centred gap ; reaching barrier ; elength constant throughout ; nly 3 wavefronts drawn max. 2, 2 drawn max 1)	1 1 1	3
		(ii)	diffra	action ;	1	1
						[Total: 7]
6	(a)	(i)	dam	ses acid rain/causes smog ; nages buildings/trees/makes breathing difficult ; nanswers must match, otherwise max 1) any two	1 1 1 + 1	2
		(ii)	•	eds up reduction of nitrogen oxide ; orm nitrogen ;	1	2
	(b)	C ₃ F	H ₈ = (3	$3 \times 12) + (8 \times 1) = 44$ and $CO_2 = 12 + (2 \times 16) = 44$;	1	
				opane produces 3 × 44 = 132 kg carbon dioxide ; opane produces 132/44 = 3.0 kg carbon dioxide ;	1 1	
				oon dioxide has volume 24dm³; arbon dioxide has volume 1000 × 3.0 × 24/44 = 1636dm³;	1 1	5
	(c)					
		one	a sh four	k each for: lared pair of electrons; r shared pairs of electrons, two for each oxygen; r other electrons on each oxygen;	1 1 1	3

[Total: 12]

	Page 5	Mark Scheme	Syllabus	Paper
		IGCSE – October/November 2008	0652	03
7	(a) cracking of an alk	; kane/oil/petroleum ;	1 1	2
		$H_2O \rightarrow C_2H_5OH ;;$ ck for each side	2	2
	(c) a catalys	st/named catalyst ;	1	1 [Total: 5]
8	(a) Use of p	power = VI 000 000/55 000	1 1	
	= 3 600		1	3
		s energy loss (in cables); me power transmitted) at lower current;	1	2
	(ii) tran	sformer;	1	1
	(iii) use	of $n_1/n_2 = V_1/V_2$; = 220 : 1;	1 1	2
	(d) energy i	nput = energy output ;	1	1
				[Total: 9]
9	(a) electron fast/ene	; rgetic/from the nucleus ;	1 1	2
	. , . ,	leon numbers correct:131 0; ton numbers correct: 54 -1;	1 1	2
	(ii) xen nob	on ; le gas ;	1 1	2
	long	rtish half life OR Xe unreactive g enough to do tests etc. but not too long to harm pat a correct sort of penetration ANY TWO	tient 1 +	1 2 [Total: 8]