MARK SCHEME for the October/November 2007 question paper

0652 PHYSICAL SCIENCE

0652/05 Paper 5 (Practical Test), maximum raw mark 30

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1 **(a)** stating the value of resistance/m should be the same as the supervisor [1]

(b)&(c) 5 values of **y** and I

good range of **y** to include 85 or more [3]

(d) **(i)** R is correctly calculated

current decreases with increasing **y**   do not allow if I greater than 1.0 [1]

(ii) IR is calculated correctly

2 places of decimals used [2]

(e) Graph

A   axes labelled

S   sensible scale used

P   plotting correct   allow one error

C   smooth curve drawn

Origin included [5]

(f) correctly read from graph must be on graph [1]

(g) explanation needs to say that current would be greater

similar curve but above experimental one [2]

[Total: 15]

2 Chemistry question

(a) **X** is colourless/no change/cloudy/stayed the same NOT clear
Y is pink
Z is pink [1]
all need to be correct

X is an acid
Y is an alkali
Z is an alkali [2]
each incorrect lose a mark to zero

(b) test correctly described TWO marks acidifying not necessary

acid is hydrochloric ONE

test can be for sulphate showing negative therefore must be chloride
it must be clear that the candidate has carried out the experiment
if in doubt, maximum of two marks [3]

(c) (i) pink colour disappears/colourless [1]

   (ii) pink colour disappears/colourless

effervescence
do not allow clear as equal to colourless [2]

(d) (i) white precipitate

      dissolves in excess [2]

   (ii) white precipitate

      precipitate does not dissolve [2]

(e) **Y** could be sodium hydroxide

Z could be sodium carbonate

a formula is allowed if it is correct [2]

[Total: 15]