



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS
International General Certificate of Secondary Education

COMBINED SCIENCE

Paper 1 Multiple Choice

0653/11

May/June 2010

45 minutes

Additional Materials: Multiple Choice Answer Sheet
Soft clean eraser
Soft pencil (type B or HB is recommended)

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Write your name, Centre number and candidate number on the Answer Sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**.

Choose the **one** you consider correct and record your choice in **soft pencil** on the separate Answer Sheet.

Read the instructions on the Answer Sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

Any rough working should be done in this booklet.

A copy of the Periodic Table is printed on page **20**.

This document consists of **17** printed pages and **3** blank pages.



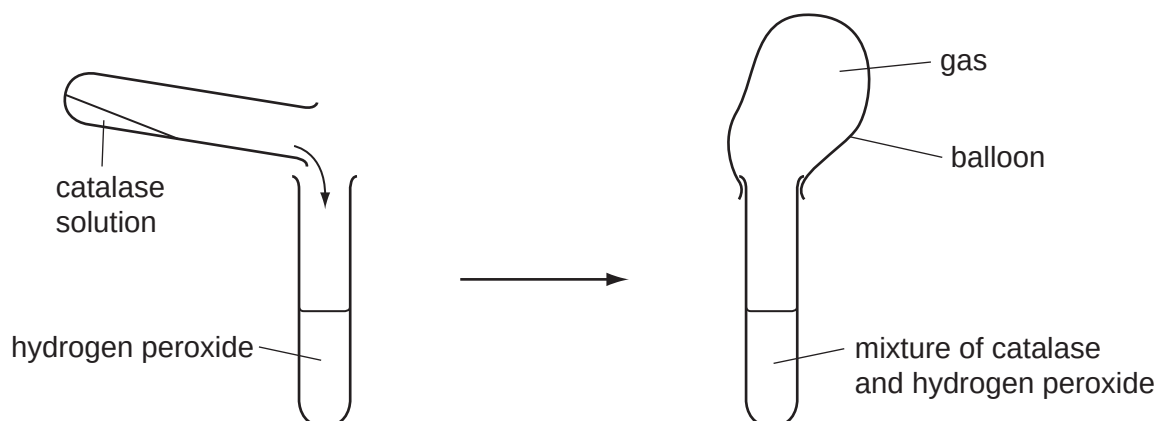
1 Which part of a plant cell controls what enters and leaves the cell?

- A cell sap
- B cell surface membrane
- C cellulose cell wall
- D cytoplasm

2 Which substances may diffuse into and out of plant cells?

	into plant cells	out of plant cells
A	chlorophyll	oxygen
B	oxygen	water
C	starch	chlorophyll
D	water	starch

3 The diagram shows a simple experiment using a solution of the enzyme catalase.

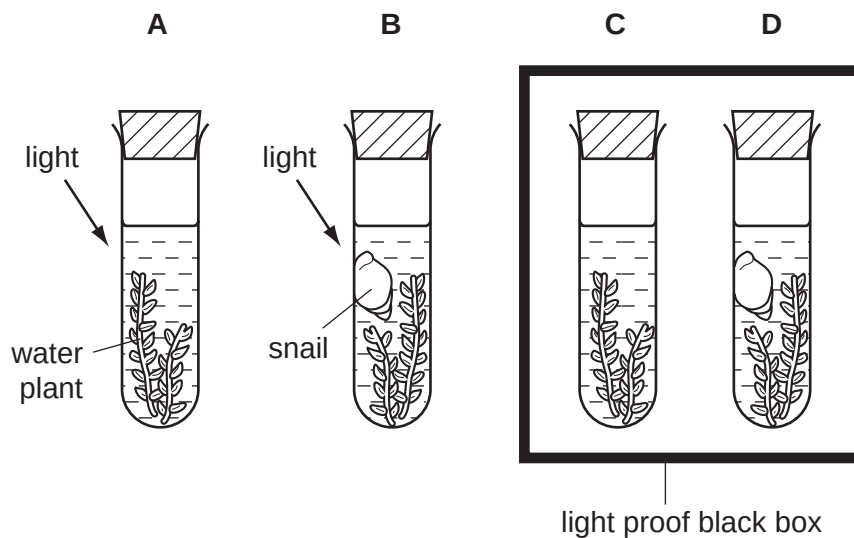


Which gas is collected in the balloon?

- A carbon dioxide
- B hydrogen
- C nitrogen
- D oxygen

- 4 The diagram shows the apparatus used in an investigation on gas exchange in organisms.

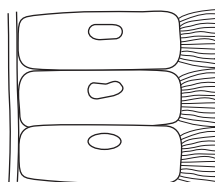
In which test-tube would the concentration of oxygen decrease most rapidly?



- 5 Which nutrients in the diet will help to prevent leg bones becoming deformed?

- A calcium and vitamin C
- B calcium and vitamin D
- C iron and vitamin C
- D iron and vitamin D

- 6 The diagram shows some cells from the lining of the human trachea.

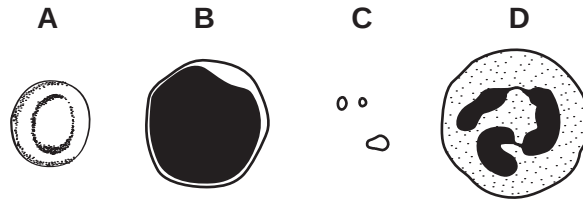


What is the function of these cells?

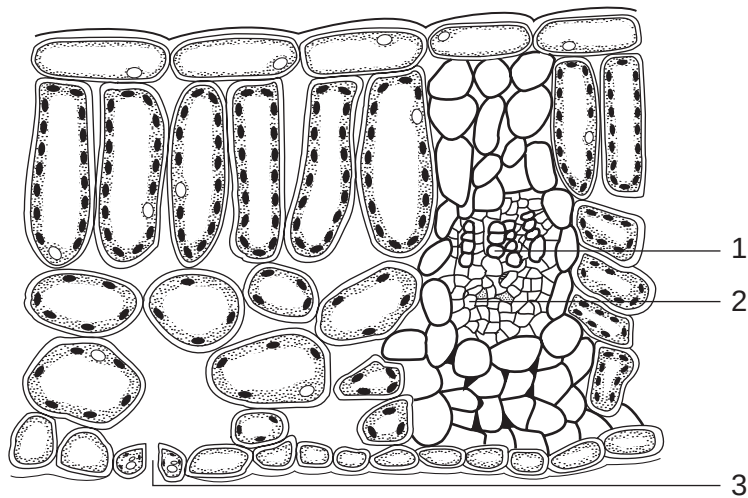
- A absorption of oxygen into the blood
- B excretion of carbon dioxide from the blood
- C removal of mucus from the trachea
- D secretion of mucus into the trachea

7 The diagram shows four components of blood.

Which component contains haemoglobin?



8 The diagram shows a section through a leaf.



Which part brings water to the leaf and through which part does water leave the leaf?

	brings water	water leaves
A	1	2
B	1	3
C	2	1
D	3	1

9 What effect does drinking alcohol have on the following:

- the rate at which signals pass along nerves,
- the length of reaction time?

	rate at which signals pass along nerves	length of reaction time
A	faster	longer
B	faster	shorter
C	slower	longer
D	slower	shorter

10 A variety of potato plant produces red tubers ('potatoes') that grow into new potato plants which then produce red 'potatoes' the following year.

Why is this?

- A** Asexual reproduction produces identical potato plants.
- B** Asexual reproduction results in different coloured 'potatoes'.
- C** Sexual reproduction produces only red 'potatoes'.
- D** Sexual reproduction requires the potato plant to produce flowers.

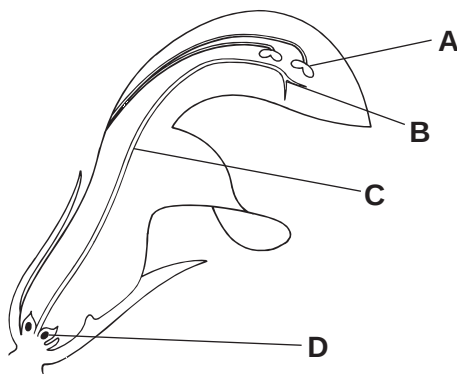
11 The table shows three ways in which different methods of birth control work.

Which correctly links the methods with how they work?

	prevents ovulation	prevents sperm release	prevents zygote implantation
A	IUD	sterilisation	cap
B	pill	rhythm	condom
C	pill	sterilisation	IUD
D	rhythm	condom	IUD

12 The diagram shows a section through a flower.

Where does fertilisation occur?

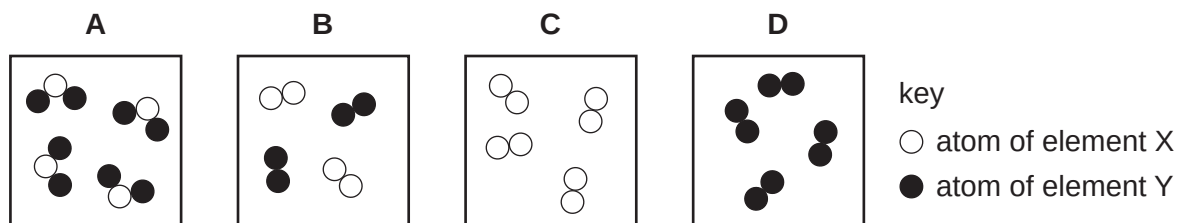


13 Rain forests have high species diversity.

What is meant by the term *species diversity*?

- A Members of a species in the forest are scattered over a wide area.
- B Members of a species show a wide range of variations.
- C There are many different species present.
- D The same species are found in many different rain forests in the world.

14 Which diagram represents a mixture of elements?

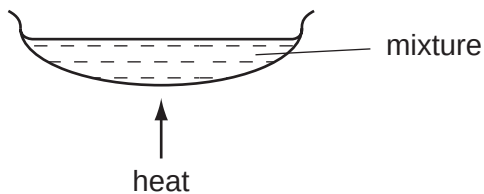


15 A solid is ionic.

Which property best confirms this fact?

- A its behaviour as an electrolyte
- B its melting point
- C its solubility in water
- D the shape of its crystals

- 16** The diagram shows apparatus used to separate a mixture.



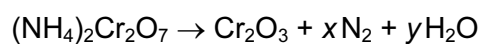
Which mixture could it be used to separate?

- A** sand and salt
B sand and sugar
C salt and sugar
D salt and water

- 17** Which letter shows the position in the Periodic Table of an unreactive non-metal?

[illegible]

- 18** The salt ammonium dichromate(VI) readily decomposes when heated.



Which values of x and y make this equation balanced?

	x	y
A	1	2
B	1	4
C	2	2
D	2	4

19 Which processes are used to purify the drinking water from reservoirs?

	chlorination	distillation	filtration
A	✓	✓	✓
B	✓	✓	x
C	✓	x	✓
D	x	✓	✓

20 Which method is used to obtain iron from iron(III) oxide?

- A** combustion
- B** electrolysis
- C** reduction
- D** thermal decomposition

21 The table gives, for two metals, the colours of their solids and their aqueous ions.

metal	colour of	
	solid	aqueous ion
copper	red-brown	blue
zinc	grey	colourless

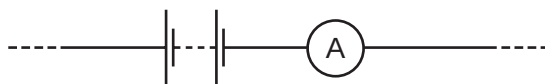
Equal amounts of powdered copper and powdered zinc are mixed together.

The mixture is added to dilute sulfuric acid in a test-tube until the acid is just used up. Some powder then remains at the bottom of the test-tube.

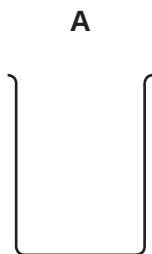
What are the colours of the remaining powder and the solution in the test-tube?

	powder	solution
A	grey	blue
B	grey	colourless
C	red-brown	blue
D	red-brown	colourless

22 Part of the apparatus used to electrolyse aqueous copper(II) chloride is shown.



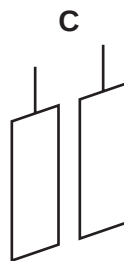
Which piece of apparatus is **not** required in this experiment?



beaker



bulb



electrodes



electrolyte

23 A student carries out experiments with zinc and dilute hydrochloric acid.

Which change in conditions makes the reaction slower?

- A** adding a suitable catalyst
- B** increasing the concentration of the acid
- C** increasing the particle size of the zinc
- D** increasing the temperature

24 What happens to a piece of magnesium when it burns in air?

- A** oxidation and combustion
- B** oxidation and thermal decomposition
- C** reduction and combustion
- D** reduction and thermal decomposition

25 This description of a plastic is incomplete.

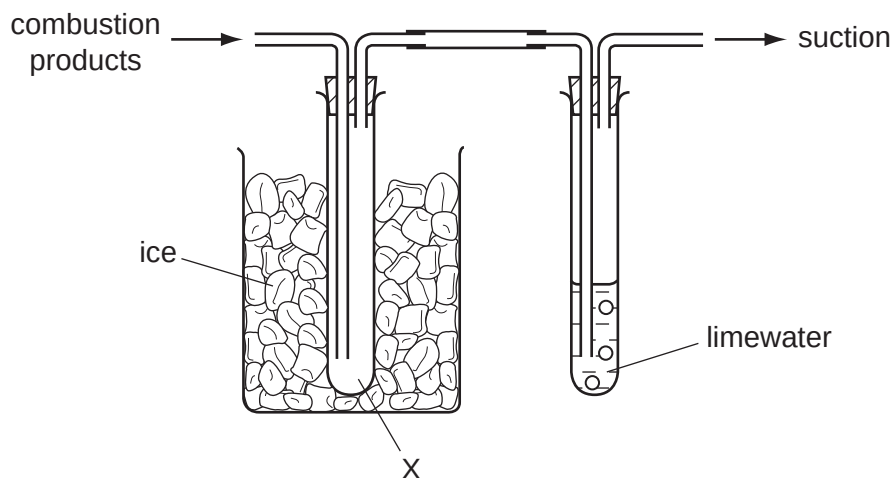
To make a plastic,1..... of a2..... combine to form a long chain3..... .

Which words correctly complete the gaps?

	1	2	3
A	atoms	monomer	polymer
B	atoms	polymer	monomer
C	molecules	monomer	polymer
D	molecules	polymer	monomer

26 A hydrocarbon is burned in excess oxygen.

The combustion products are passed through the apparatus shown.



What collects at X and what happens to the limewater?

	collects at X	limewater turns
A	soot	cloudy
B	soot	red
C	water	cloudy
D	water	red

27 Coal, hydrogen, methane and gasoline (petrol) are commonly used as fuels.

How many of these fuels are solids, liquids or gases?

	solids	liquids	gases
A	0	2	2
B	1	1	2
C	1	2	1
D	2	2	0

28 The table shows the dimensions of four rectangular blocks.

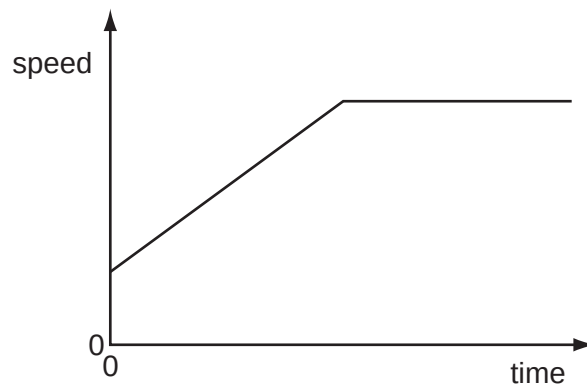
Which block has a volume of 3 m^3 ?

	height / cm	width / cm	depth / cm
A	100	100	100
B	300	100	100
C	300	300	100
D	300	300	300

29 Which statement is correct?

- A** Mass is a force.
- B** Mass is measured in newtons.
- C** Weight is a force.
- D** Weight is measured in kilograms.

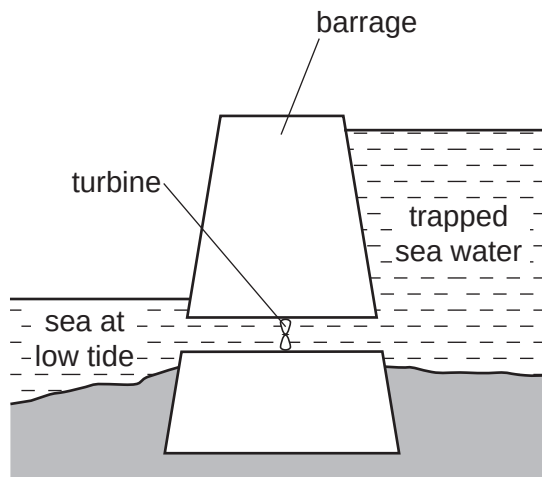
30 The data from an aeroplane flight recorder is used to plot the speed / time graph for part of the flight.



Which statement describes this part of the flight?

- A** The aeroplane accelerates from rest to a constant speed.
- B** The aeroplane decelerates after flying at a constant speed.
- C** The aeroplane reaches a constant speed after a period of changing speed.
- D** The aeroplane travels at an increasing speed followed by a decreasing speed.

- 31 In which situation would the object described be given an increase in its total energy?
- A a battery-powered torch being switched on
 - B a firework exploding
 - C a parachutist falling to the ground
 - D a skier being pulled up a slope
- 32 A tidal power station is made by building a barrage. At high tide the sea water is trapped behind the barrage.



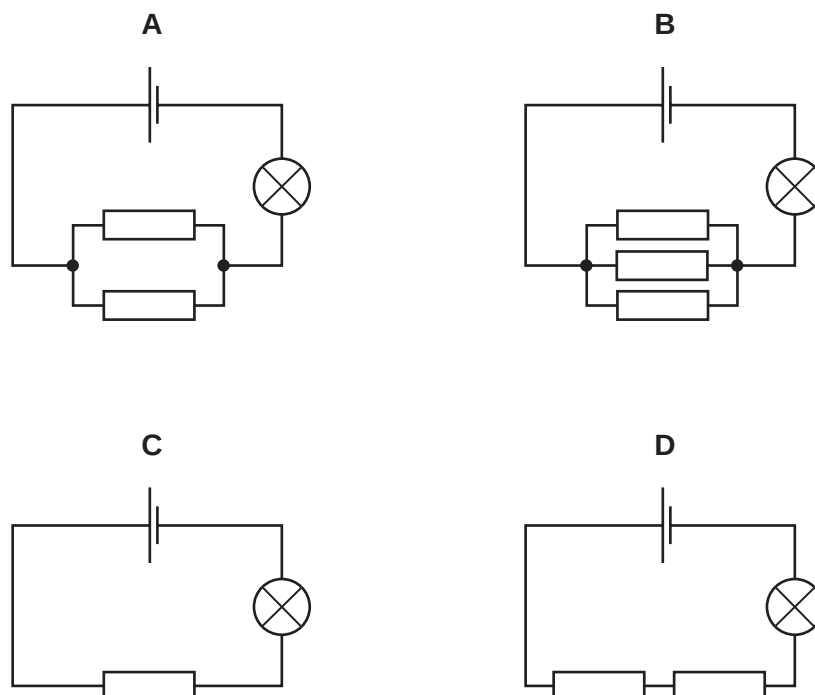
At low tide the water is allowed to flow back into the sea through a turbine.

What is the useful energy change in a tidal power station?

- A electrical energy $\rightarrow$ kinetic energy
- B electrical energy $\rightarrow$ potential energy
- C kinetic energy $\rightarrow$ potential energy
- D potential energy $\rightarrow$ electrical energy

33 In each of the circuits shown, identical resistors, cells and lamps are used.

In which circuit will the lamp glow most brightly?

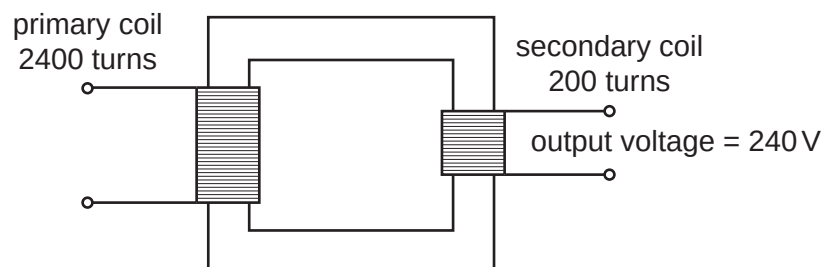


34 Electrical appliances are often fitted with a fuse.

What is the purpose of the fuse?

- A** to adjust the voltage supplied to the correct value
- B** to allow the appliance to be switched on by a small current
- C** to cut the power supply if the appliance is used too often
- D** to cut the power supply if the current becomes too large

35 A transformer has 2400 turns on its primary coil and 200 turns on its secondary coil.

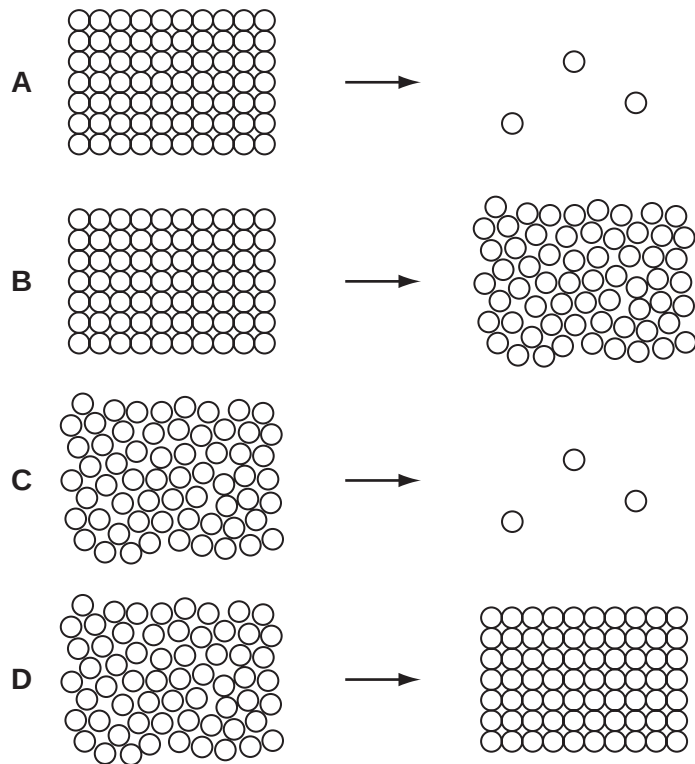


Assuming the transformer is 100 % efficient, what input voltage is needed to give an output voltage of 240V?

- A** 12V
- B** 20V
- C** 240V
- D** 2880V

36 Water spilled on the ground on a hot day evaporates.

Which diagram represents the change in arrangement of the particles in the water as it evaporates?



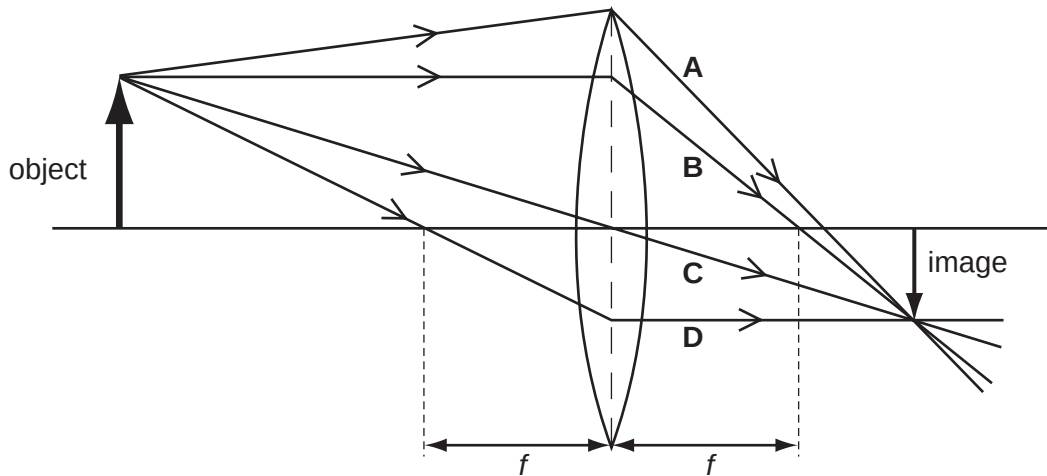
37 There is a vacuum between the double walls of a vacuum flask.

Which types of heat transfer are reduced by the vacuum?

- A** conduction and convection only
- B** conduction and radiation only
- C** convection and radiation only
- D** conduction, convection and radiation

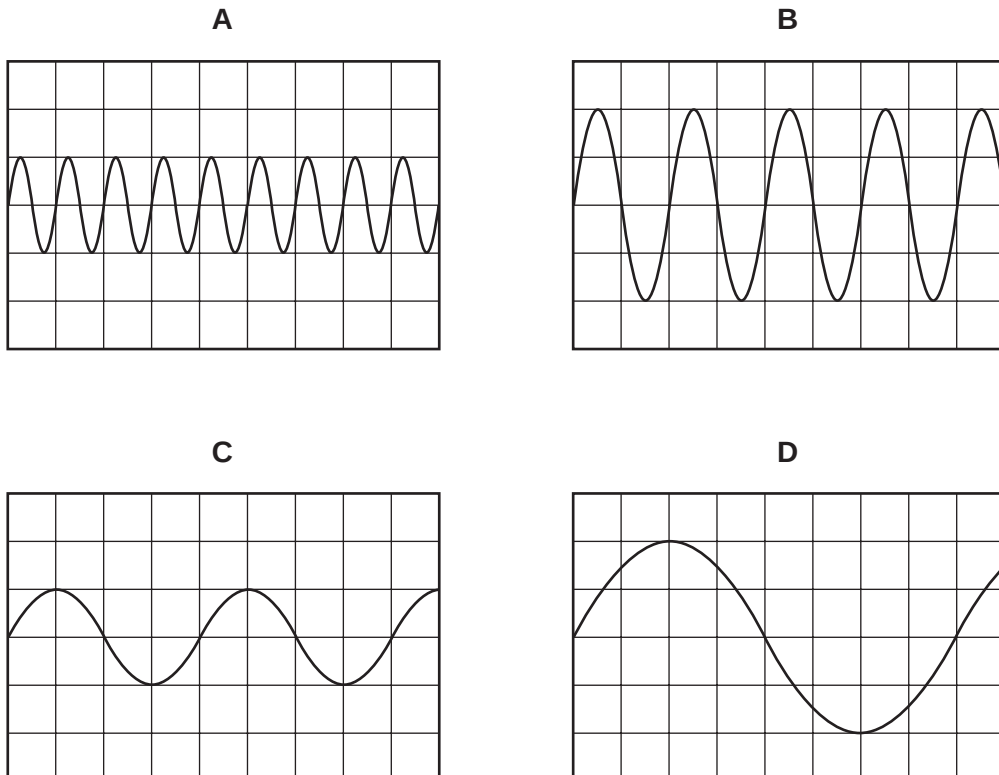
- 38 A ray diagram is drawn to locate the position of the image formed by a converging lens of focal length f .

Which ray **cannot** be drawn until the position of the image is known?

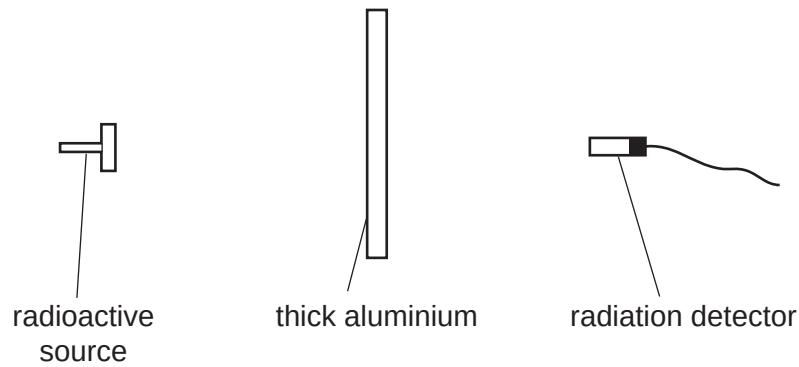


- 39 The diagrams show the wave patterns of four notes shown on a cathode ray oscilloscope. The oscilloscope controls are set the same for each sound.

Which note has the highest pitch?



40 The diagram shows a radioactive source, a thick aluminium sheet and a radiation detector.



The radiation detector shows a reading greater than the background reading.

Which type of radiation is being emitted by the source and detected by the detector?

- A** alpha-radiation
- B** beta-radiation
- C** gamma-radiation
- D** infra-red radiation

DATA SHEET
The Periodic Table of the Elements

Group																														
I	II											III	IV	V	VI	VII	0													
		1 H Hydrogen																4 He Helium												
7 Li Lithium	9 Be Beryllium											11 B Boron	12 C Carbon	14 N Nitrogen	16 O Oxygen	19 F Fluorine	20 Ne Neon													
23 Na Sodium	24 Mg Magnesium											27 Al Aluminium	28 Si Silicon	31 P Phosphorus	32 S Sulfur	35.5 Cl Chlorine	40 Ar Argon													
39 K Potassium	40 Ca Calcium	45 Sc Scandium	48 Ti Titanium	51 V Vanadium	52 Cr Chromium	55 Mn Manganese	56 Fe Iron	59 Co Cobalt	59 Ni Nickel	64 Cu Copper	65 Zn Zinc	70 Ga Gallium	73 Ge Germanium	75 As Arsenic	79 Se Selenium	80 Br Bromine	84 Kr Krypton													
85 Rb Rubidium	88 Sr Strontium	89 Y Yttrium	91 Zr Zirconium	93 Nb Niobium	96 Mo Molybdenum	101 Ru Ruthenium	101 Rh Rhodium	103 Pd Palladium	106 Ag Silver	108 Cd Cadmium	112 In Indium	115 Sn Tin	119 Sb Antimony	122 Te Tellurium	127 I Iodine	131 Xe Xenon	131 Xe Xenon													
133 Cs Caesium	137 Ba Barium	139 La Lanthanum	178 Hf Hafnium	181 Ta Tantalum	184 W Tungsten	186 Re Rhenium	190 Os Osmium	192 Ir Iridium	195 Pt Platinum	197 Au Gold	201 Hg Mercury	204 Tl Thallium	207 Pb Lead	209 Bi Bismuth	210 Po Polonium	210 At Astatine	210 Rn Radon													
87 Fr Francium	226 Ra Radium	227 Ac Actinium																												
*58-71 Lanthanoid series																														
†90-103 Actinoid series																														
140 Ce Cerium																		141 Pr Praseodymium	144 Nd Neodymium	150 Sm Samarium	152 Eu Europium	157 Gd Gadolinium	159 Tb Terbium	162 Dy Dysprosium	165 Ho Holmium	167 Er Erbium	169 Tm Thulium	173 Yb Ytterbium	175 Lu Lutetium	
232 Th Thorium																		238 Pa Protactinium	238 U Uranium	238 Np Neptunium	238 Pu Plutonium	238 Am Americium	238 Cm Curium	238 Bk Berkelium	238 Cf Californium	238 Es Einsteinium	238 Fm Fermium	238 Md Mendelevium	238 No Nobelium	238 Lr Lawrencium
Key	X	a	a = relative atomic mass																											
			X = atomic symbol																											
		b	b = proton (atomic) number																											

The volume of one mole of any gas is 24 dm³ at room temperature and pressure (r.t.p.).

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