## MARK SCHEME for the October/November 2008 question paper

## **0620 CHEMISTRY**

0620/05

Paper 5 (Practical Test), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Mark Scheme Syllabus	Paper
	IGCSE – October/November 2008 0620	05
Expe	ent 5 observations	
-	porous o.w.t.t.e (1) bubbles/fizz (1) cipitate (1)	[max 2]
initial final differ	results xes correctly completed (1) tes correctly completed (1) ces correctly completed (1) signs correct (1) able to Supervisor's results (1)	[5]
	bars correctly drawn (3), –1 for each incorrect elled (1)	[4]
(b)	solid <b>A</b> /Experiment 1 (1)	[1]
(	temperature increased/heat/energy given out (1)	[1]
<b>(c)</b> E	eriment 3/solid <b>C</b> or from results (1)	[1]
(d)	double the value or 2 × value from table (1) e.c.f. applies	[1]
(	half the value or half the value from table (1)	[1]
(i	more/larger volume of water (1) twice as much (for solid to dissolve in) of	o.w.t.t.e. (1) [2]
(e) a	l present (1) carbonate present (1) carbon dioxide (1)	[max 2]
		[Total: 20]
	ition K blue/green ition L yellow/orange/brown (1)	[1]
F	ck pHs from Supervisor's results of solution <b>K</b> 1–6 of solution <b>L</b> 1–6 (1)	[1]
tests	solution K	
(c)	blue (1) precipitate (1)	[2]
	deep/royal blue (1) solution/precipitate dissolves (1)	[2]
(		
	no reaction/change/nothing (1) not lighter	[1]

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE – October/November 2008	0620	05
tests on solut	ion L		
(d) (i) (rust	y) brown (1) precipitate (1)		[2
(ii) brov	n precipitate (1) not red		[
(iii) no re	eaction/change/nothing (1)		[
(iv) white	e (1) precipitate (1) not lighter/paler		[2
<b>(e)</b> acidic (1	)		[
(f) chloride/	C <i>l</i> <sup>-</sup> (1)		[
( <b>g) (i)</b> copp	per (1)		[
(ii) iron	(1) (III) (1)		[;
			[Total: 20
		[Total	for paper: 40