

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

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CENTRE NUMBER	CANDIDATE NUMBER		
CANDIDATE NAME			

Paper 6 Alternative to Practical

May/June 2011

1 hour

Candidates answer on the Question Paper.

Additional Materials: ruler

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO **NOT** WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use	
1	
2	
3	
Total	

This document consists of **9** printed pages and **3** blank pages.



1 Some students carried out tests for vitamin C.

They were provided with three vitamin C solutions, **S1**, **S2** and **S3**.

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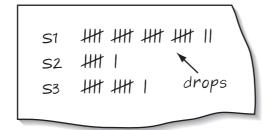
S1 had a concentration of 0.2% vitamin C.

\$2 had a concentration of 0.05% vitamin C.

The concentration of **S3** was not known.

- The students measured 1 cm³ of starch solution into a test-tube.
- They added 1 cm³ of solution S1.
- The students added iodine solution, counting drop by drop, until a blue colour appeared. This was the end-point for solution **S1**.
- They repeated the test on solutions **S2** and **S3**.

These are the results that the students recorded.



(a) Record the students' observations in a suitable table using the space below.

[4]

(b)	Use these results to suggest the approximate vitamin C concentration of S3.
	Give reasons for your answer.
	[3]
(c)	Suggest four ways in which you could improve this method to find the concentration of an unknown vitamin C solution.
	1.
	2.
	3.
	4.
	[4]

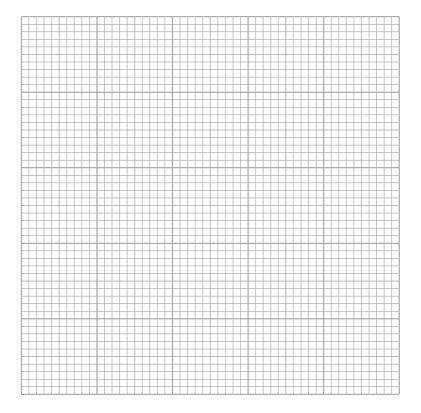
(d) Fig. 1.1 shows the results of a similar investigation into the concentration of vitamin C in five fruit juices. The students counted the number of drops of iodine solution used to reach the end-point for each fruit juice.

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Blackcurrant 48 Pineapple 5
Orange 16 Lemon 12
Strawberry 22

Fig. 1.1

(i) On the grid below plot the data from Fig. 1.1 to show the variation in the number of drops of iodine solution required to reach the end-point.



[5]

(ii) State which fruit juice has the highest concentration of vitamin C.

[1]

[Total: 17]

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Question 2 begins on Page 6

2 Fig. 2.1 shows a photograph of the larva of an insect.

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Fig. 2.1

(a) (i) In the space below make a large drawing of the larva shown in Fig. 2.1.Labels are **not** needed.

[5]

(ii)	Measure the length of the larva in Fig. 2.1 and in your drawing.		
	length of larva in Fig. 2.1		
	length of larva in your drawing		[2]
(iii)	Calculate the magnification of yo	our drawing compared with the larva in Fig. 2.1.	
	Show your working.		
		magnification	[2]

(b) The larva eats through leaf tissue making tunnels in which it lives.

Fig 2.2 shows part of a leaf that has been damaged by these tunnels.

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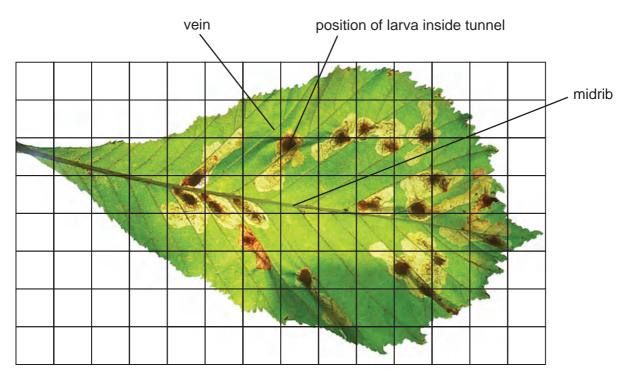


Fig. 2.2

(i) Calculate the percentage of the leaf area which has been damaged by the tunnels.Show your working.

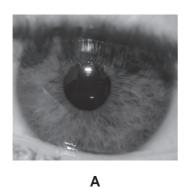
	answer	% [3]
(ii)	Suggest and explain why the tunnels do not extend across the leaf midrib.	
		[2]

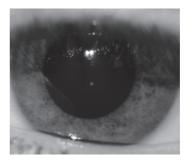
(iii)	Suggest two reasons why the leaf in Fig. 2.2 may die and fall off.
	1.
	2.
	[2]
(c) The	larva in Fig. 2.1 becomes a moth.
Fig.	2.3 and Fig. 2.4 show the moth.
1	
	Fig. 2.3 Fig. 2.4
Loo	k at Fig. 2.3 and Fig. 2.4.
(i)	State one visible feature of this moth which is used to classify it as an arthropod.
	[1]
(ii)	State three visible features of this moth which are used to classify it as an insect.
	1
	2
	3[3]
	[Total: 20]

3 Fig. 3.1 shows two photographs of a person's eye.

In photograph **A** the person was looking out of a window.

In photograph **B** the person had turned away from the window.





В

Fig 3.1

(a)	Describe what happened to the diameter of the pupil in photograph B .	
		[1]
(b)	Explain your observation.	
		[2]

[Total: 3]

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