UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

GCE Advanced Subsidiary Level

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

8780 PHYSICAL SCIENCE

8780/04

Paper 4, maximum raw mark 30

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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MMO (a) (ii) 0.1 cm or 0.2 cm [1] **(b)** 6 different sets of readings [1] range: at least from 25.0 – 10.0 cm, no gaps greater than 3.5 cm unless skewed for ℓ^2 [1] quality: h increases consistently as ℓ increases (check from graph) [1] **PDO (b)** all columns headed, ℓ /cm, h/cm, ℓ 2 /cm 2 [1] all raw data in a column to same precision [1] the calculated column correct and ℓ^2 to same number, or 1 more, significant figures as ℓ [1] **ACE** (c) (i) axes labelled, sensible scales chosen (at least half graph paper used and no awkward scales) [1] 6 points plotted correctly [1] (ii) best-fit straight line or curve for candidate's data [1] (iii) attempt to measure gradient using at least ½ range of data and dy/dx / tangent drawn if necessary [1] correct calculation of gradient [1] (d) (i) suitable limitation: e.g. parallax error in reading h or oscillation [1] (ii) suitable improvement, e.g. clamping ruler, repeat AND calculate average [1] (iii) suitable argument, consistent with graph [1] (acceptable answers: no because it's a curve; no because the straight line does not go through the origin; yes because it's a straight line through the origin)

[Total: 15]

<u>га</u>	ge 3	Mark Scheme. Teachers Version	Syllabus	Fapei
		GCE AS LEVEL – October/November 2011	8780	04
(a)	yellow w	OO /does not relight AND cloudy/milky/chalky/white precipita hen hot/during heating AND en cold/cools down	ate	[1] [1]
(b)	accuracy	perature readings to 0.1/0.05 °C		[1] [1]
(c)		ecipitate AND on/no precipitate		[1]
(d)	white pre	ecipitate with ammonia ecipitate with sodium hydroxide in excess in BOTH		[1] [1] [1]
(e)		onclusions must be consistent with the observations rec	orded)	[1] [1]
(f)	correct u	alculation of Q se of mass and molar mass se of units		[1] [1] [1]
(g)		source of error / valid improvement ion of how it affects enthalpy change		[1] [1]

Mark Scheme: Teachers' version

Syllabus

Paper

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