

**UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**Specimen for 2007**

**GCE A LEVEL**

**MARK SCHEME**

**MAXIMUM MARK: 30**

**SYLLABUS/COMPONENT: 9701/05**

**CHEMISTRY  
PRACTICAL**



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Skill	Total marks	Approximate breakdown of marks		Question 1	Question 2
Planning	15 marks	Defining the problem	4 marks	4	0
		Methods	11 marks	11	0
Analysis, conclusions and evaluation	15 marks	Dealing with data	8 marks	0	8
		Evaluation	4 marks	0	4
		Conclusion	3 marks	0	3

PLAN = Planning  
         Problem = Defining the problem  
         Methods  
 ACE = Analysis, conclusions and evaluation  
         Data = Dealing with data  
         Evaluation  
         Conclusions

Question	Sections	Learning outcomes	Indicative material	mark
<b>1</b>	<b>(a) (i)</b>	PLAN Problem	• identify the independent variable in the experiment or investigation	temperature and size of marble chips <b>2</b>
	<b>(ii)</b>		• identify the dependent variable in the experiment or investigation	volume or mass of CO <sub>2</sub> <b>1</b>
	<b>(b)</b>	PLAN Problem	• express the aim in terms of a prediction or hypothesis, and express this in words or in the form of a predicted graph	suitable hypothesis proposed e.g. rate of production of CO <sub>2</sub> increases with increasing concentration of hydrochloric acid. <b>1</b>

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	(c)	PLAN Methods	<ul style="list-style-type: none"> <li>describe how the dependent variable is to be measured</li> <li>describe the arrangement of apparatus and the steps in the procedure to be followed</li> <li>describe the method to be used to vary the independent variable, and the means to ensure that its value is measured accurately</li> <li>describe how each of the other key variables is to be controlled</li> <li>describe precautions that should be taken to keep risks to a minimum</li> <li>suggest appropriate volumes and concentrations of reagents</li> </ul>	<p>appropriate apparatus to measure volume or mass of CO<sub>2</sub>;</p> <p>diagram showing appropriate apparatus and stepwise description including time measurement</p> <p>appropriate volumes of acid and water;</p> <p>use of appropriate apparatus in measuring volumes of acid and water;</p> <p>control of temp and constant number and size of marble chips (e.g. same mass and number of chips)</p> <p>care when making up HCl from conc. HCl</p> <p>moles/mass of CaCO<sub>3</sub> calculated;</p> <p>initial [HCl] calculated</p>	<p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>1</p> <p>9 max 8</p>
	(d)	PLAN Methods	<ul style="list-style-type: none"> <li>draw up tables for data that they might wish to record</li> <li>describe how the data might be used in order to reach a conclusion</li> </ul>	<p>columns for mass/concentration/time</p> <p>units correct</p> <p>calculation of CO<sub>2</sub> evolved, appropriate statement relating to hypothesis in (b)</p>	<p>1</p> <p>1</p> <p>1</p>
<b>1 Total</b>					<b>15</b>



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	(f)	ACE Conclusions	<ul style="list-style-type: none"> <li>draw conclusions from an investigation, providing a detailed description of the key features of the data and analyses, and considering whether experimental data supports a given hypothesis</li> </ul>	makes appropriate comment on whether prediction is supported by data i.e. straight line graph	1
		ACE Evaluation	<ul style="list-style-type: none"> <li>make informed judgements on the confidence with which conclusions may be drawn</li> </ul>	makes appropriate comment on whether procedure is suitable for determination of $M_r$	1
	(g)	ACE Conclusions	<ul style="list-style-type: none"> <li>make further predictions, ask informed and relevant questions and suggest improvements</li> </ul>	suggests appropriate modification to experimental procedure such as more points in range where accuracy is greatest	1
	(h)	ACE Conclusions	<ul style="list-style-type: none"> <li>make detailed scientific explanations of the data, analysis and conclusions that they have described</li> <li>make further predictions, ask informed and relevant questions and suggest improvements</li> </ul>	uses knowledge of acid/base chemistry to describe a more appropriate way of determining $M_r$ such as titration.	1
<b>2 Total</b>					<b>15</b>