## **UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

Specimen for 2007

## **GCE A LEVEL**

## MARK SCHEME

**MAXIMUM MARK: 30** 

SYLLABUS/COMPONENT: 9701/05

CHEMISTRY PRACTICAL

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Skill	Total marks	Approximate breakdown of marks		Question 1	Question 2
Planning	15 marks	Defining the problem	4	4	0
			marks		
		Methods	11	11	0
			marks		
Analysis,	15 marks	Dealing with data	8	0	8
conclusions			marks		
and evaluation		Evaluation	4	0	4
			marks		
		Conclusion	3	0	3
			marks		

PLAN = Planning

Problem = Defining the problem

Methods

ACE = Analysis, conclusions and evaluation

Data = Dealing with data

Evaluation Conclusions

Que	estion	Sections	Learning outcomes	Indicative material	mark
1	(a) (i) (ii)	PLAN Problem	<ul> <li>identify the independent variable in the experiment or investigation</li> <li>identify the dependent variable in the experiment or investigation</li> </ul>	temperature and size of marble chips  volume or mass of CO <sub>2</sub>	1
	(b)	PLAN Problem	express the aim in terms of a prediction or hypothesis, and express this in words or in the form of a predicted graph	suitable hypothesis proposed e.g. rate of production of CO <sub>2</sub> increases with increasing concentration of hydrochloric acid.	1

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1 Tc	atal			, pour.out (0)	15
	(d)	PLAN Methods	<ul> <li>draw up tables for data that they might wish to record</li> <li>describe how the data might be used in order to reach a conclusion</li> </ul>	columns for mass/ concentration/time  units correct  calculation of CO <sub>2</sub> evolved, appropriate statement relating to hypothesis in <b>(b)</b>	1 1 1
			reagents	initial [HC <i>I</i> ] calculated	9 max 8
			<ul> <li>suggest appropriate volumes and concentrations of</li> </ul>	moles/mass of CaCO <sub>3</sub> calculated;	1
			<ul> <li>describe precautions that should be taken to keep risks to a minimum</li> </ul>	size of marble chips (e.g. same mass and number of chips)  care when making up HCl from conc. HCl	1
			<ul> <li>describe how each of the other key variables is to be controlled</li> </ul>	water; control of temp and constant number and	1
			<ul> <li>describe the method to be used to vary the independent variable, and the means to ensure that they its value is measured accurately</li> </ul>	appropriate volumes of acid and water; use of appropriate apparatus in measuring volumes of acid and	1
			<ul> <li>describe the arrangement of apparatus and the steps in the procedure to be followed</li> </ul>	diagram showing appropriate apparatus and stepwise description including time measurement	1 1
	(c)	PLAN Methods	describe how the dependent variable is to be measured	appropriate apparatus to measure volume or mass of CO <sub>2</sub> ;	1

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2	(a)	ACE Data	identify the calculations that are necessary to be able to draw conclusions from provided data	calculates mass of $X_2CO_3$ used and mass of $CO_2$ evolved;	2
	(b)	ACE Data	use graphs to draw attention to the key points in quantitative data, including the variability of data	plots mass of CO <sub>2</sub> on y-axis, mass of X <sub>2</sub> CO <sub>3</sub> on x-axis with appropriate labels and units; suitable scales – points plotted over more than half of each axis; correct plotting of at least 10 points; appropriate best-fit line drawn	1 1 1
	(c)	ACE Evaluation	<ul> <li>identify anomalous values in provided data and suggest appropriate means of dealing with such anomalies</li> <li>within familiar contexts, suggest possible explanations for anomalous readings</li> </ul>	identifies one point where too much CO <sub>2</sub> produced – (cotton wool plug not weighed at end);  identifies one point where too little CO <sub>2</sub> produced – (solution not saturated with CO <sub>2</sub> at start, or not left for 10 mins for CO <sub>2</sub> to diffuse)	1
	(d)	ACE Evaluation	identify the extent to which provided readings have been adequately replicated, and describe the adequacy of the range of data provided	identifies less reliability with lower masses of X <sub>2</sub> CO <sub>3</sub>	1
	(e)	ACE Data	use calculations to enable simplification or explanation of data	values read from graph. NOT table values; calculates $M_r$	1

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(f)	ACE Conclusions  ACE Evaluation	<ul> <li>draw conclusions from an investigation, providing a detailed description of the key features of the data and analyses, and considering whether experimental data supports a given hypothesis</li> <li>make informed judgements on the confidence with which conclusions may be drawn</li> </ul>	makes appropriate comment on whether prediction is supported by data i.e. straight line graph  makes appropriate comment on whether procedure is suitable for determination of <i>M</i> <sub>r</sub>	1
(g)	ACE Conclusions	make further predictions, ask informed and relevant questions and suggest improvements	suggests appropriate modification to experimental procedure such as more points in range where accuracy is greatest	1
(h)	ACE Conclusions	<ul> <li>make detailed scientific explanations of the data, analysis and conclusions that they have described</li> <li>make further predictions, ask informed and relevant questions and suggest improvements</li> </ul>	uses knowledge of acid/base chemistry to describe a more appropriate way of determining $M_r$ such as titration.	1
2 Total				15