

**UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**GCE Advanced Subsidiary Level and GCE Advanced Level**

## **MARK SCHEME for the May/June 2007 question paper**

### **9701 CHEMISTRY**

**9701/31**

Paper 31 (Advanced Practical Skills 1),  
maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

- CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2007 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



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**Generic Mark Scheme for Papers 31 and 32**

| <b>Skill</b>                              |          | <b>Breakdown of marks</b>  |         |
|---|----------|--|---------|
| Manipulation, measurement and observation | 16 marks | Successful <u>collection</u> of data and observations                          | 8 marks |
|   |          | <u>Decisions</u> relating to measurements or observations                      | 8 marks |
| Presentation of data and observations     | 12 marks | <u>Recording</u> data and observations   | 5 marks |
|   |          | <u>Display</u> of calculation and reasoning                                    | 3 marks |
|   |          | Data <u>layout</u>   | 4 marks |
| Analysis, conclusions and evaluation      | 12 marks | <u>Interpretation</u> of data or observations and identifying sources of error | 6 marks |
|   |          | Drawing <u>conclusions</u>   | 5 marks |
|   |          | Suggesting <u>improvements</u>   | 1 mark  |

|        |                                |          |       |
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| Page 3 | Mark Scheme                    | Syllabus | Paper |
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| Question         | Sections                | Indicative material  | Mark           |     |
|------------------|-------------------------|--|----------------|-----|
| 1 (a)            | (i) PDO<br>Layout       | Tabulates initial and final burette readings and volume added in each of the tables<br><i>Tabulation may be vertical or horizontal.</i><br><i>Ignore absence of units</i><br><i>Do NOT award this mark if any final and initial burette readings are inverted or 50 is used as the initial burette reading</i>   | [1]            |     |
|                  | (ii) PDO<br>Recording   | Both burette readings in the dilution table and <u>final and initial</u> burette readings for all accurate titres in the titration table recorded to the nearest 0.05 cm <sup>3</sup> .<br><i>Treat all titres as “accurate” unless labelled rough or 1st titre is to lower precision than subsequent titres</i>   | [1]            |     |
|                  | (iii) MMO<br>Collection | Follows instructions – Rough plus sufficient accurate titrations<br><i>Award this mark if there are three or more titres <u>OR where two titres only have been recorded</u> they are within 0.20 cm<sup>3</sup> (neither labelled as rough). The first titre does not have to be labelled rough</i>  | [1]            |     |
|                  | (iv) MMO<br>Decisions   | Has at least two uncorrected titres within 0.1 cm <sup>3</sup><br><br><b>Accuracy (v) and (vi)</b><br>Give 2 marks if difference to Supervisor is <b>0.3</b> or less<br>Give 1 of these two marks for a difference of <b>0.3+ to 0.5</b><br>Give 0 marks for a difference greater than <b>0.5</b>  | [1]<br><br>[2] | [6] |
| (b)              | ACE<br>Interpretation   | Candidate selects/calculates appropriate “average” from any uncorrected titre values within 0.20 cm <sup>3</sup> .   | [1]            | [1] |
| (c) (i) and (ii) | ACE<br>Interpretation   | Examiner checks each of the first four steps of the calculation.<br>Award two marks if all steps are chemically correct.<br>Withhold 1 mark for each chemical error – no negative marks. Count non-completed steps as chemical errors.<br><br>$\frac{\text{vol diluted}}{1000} \times 0.50$ $\times \frac{\text{titre}}{250}$ $\times \frac{1}{2} \times \frac{1000}{25} \quad (\text{Potential 2 errors})$ $\times 106$ | [2]            |     |
|                  | (iii) PDO<br>Display    | Working shown in each step attempted   | [1]            |     |
|                  | (iv)                    | 3 or 4 significant figures in final answer given for each of the first four steps  | [1]            |     |
|                  | (v)                     | Answer to last section is correctly evaluated to 3 sf using candidate’s value to 4 <sup>th</sup> step.<br><i>(Answer may be from final answer to step 4 or use number carried on calculator)</i>   | [1]            | [5] |

|        |                                |          |       |
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| Question | Sections           | Indicative material   | Mark        |     |
|----------|--------------------|---|-------------|-----|
| (d)      | ACE Interpretation | Smallest division correctly read from measuring cylinder and error estimated at $\frac{1}{2}$ smallest division   | [1]         | [2] |
|          |                    | Both % errors correctly calculated<br><i>[Award second mark ecf from smallest division and estimated error]</i>   | [1]         |     |
| (e)      | ACE Conclusions    | Draws appropriate conclusion from (d) – supported by (experimental) evidence <ul style="list-style-type: none"> <li>• compares consistency of two titres performed</li> <li>• comments on lower titre than in 1<sup>st</sup> experiment</li> <li>• compares % error for measuring cylinder and pipette</li> <li>• refers to liquid remaining in the measuring cylinder (allowed for in pipette)</li> <li>• measuring cylinder has large error – volumes added will vary (each time) leading to variable titres</li> </ul> | [1]         | [1] |
| (f)      | ACE Improvement    | Candidate suggests heating solution to drive off CO <sub>2</sub><br><i>(accept use of hot water only if linked to decreased solubility of carbon dioxide)</i>   | [1]         | [1] |
|          |                    |   | [Total: 16] |     |

|        |                                |          |       |
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| Question | Sections              | Indicative material   | Mark       |     |
|----------|-----------------------|---|------------|-----|
| 2 (a)    | PDO<br>Layout         | Tabulates all experimental readings for at least one experiment.<br>(mass of empty weighing bottle, mass of bottle + solid, mass of bottle + residual solid, initial temperature, final temperature)<br>and $\Delta T$ and $m$                            | [1]        | [4] |
|          | PDO<br>Recording      | Tabulated values are in a single table covering both experiments<br><br>Table has correct labels and units ( <u>only g and °C</u> )   | [1]<br>[1] |     |
|          | MMO<br>Collection     | All weighings recorded with consistent precision to at least 1 dp and all temperature readings recorded to 1 dp only.   | [1]        |     |
| (b)      | MMO<br>Collection     | Give one mark if difference between candidate's $\Delta T/m$ values, is within $0.1\text{ }^{\circ}\text{C g}^{-1}$   | [1]        | [2] |
|          |                       | Give one mark if the difference between mean $\Delta T/m$ value for Supervisor and closer $\Delta T/m$ value of candidate, is within $0.1\text{ }^{\circ}\text{C g}^{-1}$   | [1]        |     |
| (c)      | MMO<br>Decisions      | Candidate <u>refers to</u> his/her experimental values to arrive at an appropriate comment as to whether the experiment should be further repeated<br><i>[The answer must be based on the reliability (consistency) of the two experiments performed]</i> | [1]        | [1] |
| (d)      | ACE<br>Interpretation | Examiner calculates to 3sf the mean $\Delta T/m$ value from the candidate's $\Delta T/m$ values for each experiment.<br>Give 1 mark if the candidate's answer to (d) is within 1% of examiner calculated mean $\Delta T/m \times 61.59$ .                 | [1]        | [1] |
|          |                       |   | [Total: 8] |     |

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| Page 6 | Mark Scheme                    | Syllabus | Paper |
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| Question   | Sections   | Indicative material  | Mark |            |
|--|--|--|------|------------|
| <b>FA 6 is solid basic zinc carbonate, FA 7 is solid copper(II) chloride</b> |  |  |      |            |
| <b>3 (a)</b>   | PDO<br>Layout  | Tabulates observations for <b>FA 6</b> and <b>FA 7</b>   | [1]  | <b>[4]</b> |
|  | MMO<br>Collection  | Observes and records blue or green solution with <b>FA 7</b> <b>and</b> colourless solution with <b>FA 6</b>   | [1]  |            |
|  | MMO<br>Decisions   | Selects lime water in test for gas with <b>FA 6</b>  | [1]  |            |
|  | ACE<br>Conclusion  | Identifies carbon dioxide as gas evolved from test with limewater<br><i>Evidence for CO<sub>2</sub> may also be found in the conclusion for the anion and can be awarded retrospectively</i>   | [1]  |            |
| <b>(b)</b>   | PDO<br>Layout  | Tabulates observations<br><i>This table should show clearly rows/columns for NaOH and NH<sub>3</sub> as reagents and <b>FA 6/FA 7</b>.<br/>The table does not need lines to be drawn – clearly laid out and headed blocks of text are acceptable</i> | [1]  | <b>[6]</b> |
|  | PDO<br>Recording   | All observations in a single table<br><i>The key feature to look for is reagent information – it should only appear once for each reagent and cover <b>FA 6</b> and <b>FA7</b></i>   | [1]  |            |
|  |  | Full observations for reagents to excess   | [1]  |            |
|  | MMO<br>Collection  | Records white ppt soluble in excess NaOH and excess NH <sub>3</sub> for <b>FA 6</b>  | [1]  |            |
|  |  | For <b>FA 7</b> records a blue ppt with NaOH and a <u>lighter/paler</u> blue ppt with ammonia  | [1]  |            |
|  | For <b>FA 7</b> records (blue/white only) ppt soluble in excess ammonia to form a deep/dark blue <u>solution</u> . | [1]  |      |            |

|        |                                |          |       |
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| Question | Sections           | Indicative material   | Mark                      |            |
|----------|--------------------|---|---------------------------|------------|
| (c)      | ACE<br>Conclusions | Cation in <b>FA 6</b><br><b>Summarises evidence for Zn<sup>2+</sup> from solubility of white ppt in NH<sub>3</sub>(aq)</b>  | [1]                       | <b>[3]</b> |
|          |                    | <i>Mark the conclusions for FA 6 consequentially for incorrect observations.</i>  |                           |            |
|          |                    | Cation in <b>FA 7</b><br>Summarises evidence for Cu <sup>2+</sup> from blue ppt with NaOH(aq) and NH <sub>3</sub> (aq) or <u>deep/dark blue</u> colour of solution with excess NH <sub>3</sub> (aq)   | [1]                       |            |
|          |                    | Anion in <b>FA 6</b><br>Summarises evidence for CO <sub>3</sub> <sup>2-</sup> from <b>(a)</b> .   | [1]                       |            |
| (d)      | MMO<br>Decisions   | <i>Expected observations are required</i><br><br>Test 1<br><br>Selects AgNO <sub>3</sub> or other soluble silver salt to test for chloride<br>Addition of Ag <sup>+</sup> (aq) or a solution containing Ag <sup>+</sup> or silver(I) ions is acceptable<br><br>Selects aqueous ammonia – added to ppt with NH <sub>3</sub><br><br>Test 2<br><br>Selects soluble lead salt e.g. Pb(NO <sub>3</sub> ) <sub>2</sub> as reagent<br>Addition of Pb <sup>2+</sup> (aq) or a solution containing Pb <sup>2+</sup> or lead(II) ions is acceptable<br><br>Give one of the last two marks if Ag <sup>+</sup> and Pb <sup>2+</sup> only are given. | [1]<br><br>[1]<br><br>[1] | <b>[3]</b> |
|          |                    |   | <b>[Total: 16]</b>        |            |